

C h e m i s t r y 1 1

# Atomic Theory

## The Rutherford Model

To account for Rutherford's discovery of the nucleus, the Rutherford model was developed. This model shows a small, positively charged ball at the center of the atom, with most of the mass concentrated there.

Notes

### Activity 1.1: Model

Students will develop a model of the atom that is consistent with the experimental evidence.



Model 1: A simple model of an atom with a central nucleus and three electrons.

Model 2:

Model 3: A model of an atom with a central nucleus and three electrons, showing the relative sizes of the nucleus and electrons.

Model 4:

### Model

Students will develop a model of the atom that is consistent with the experimental evidence.

Model 1:



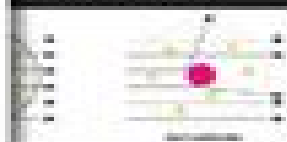
Model 2: A model of an atom with a central nucleus and three electrons, showing the relative sizes of the nucleus and electrons.

Model 3: A model of an atom with a central nucleus and three electrons, showing the relative sizes of the nucleus and electrons.



### Hydrogen Atom

Model 1: A model of a hydrogen atom with a central nucleus and one electron.



The  
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Scientist

# Fill In Blank Questions On Atomic Theory Df

**Lingjun Ying**



**Fill In Blank Questions On Atomic Theory Df:**

## **Fill In Blank Questions On Atomic Theory Df** Book Review: Unveiling the Magic of Language

In an electronic digital era where connections and knowledge reign supreme, the enchanting power of language has become apparent than ever. Its capability to stir emotions, provoke thought, and instigate transformation is really remarkable. This extraordinary book, aptly titled "**Fill In Blank Questions On Atomic Theory Df**," compiled by a highly acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound affect on our existence. Throughout this critique, we shall delve into the book's central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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