Equilibrium Practice Test

1012101102200000	Name: ANSWER	ES	60 /60
witiple Choice cutify the letter of the choice that i	to the second second second	most or our service the	assertion (20 months)
	And the same of the same of		desired 15 of section 17 h
1. Equilibrium hoppens when			
a, opposing forces belance b. opposing forces are unbal			rise reactions occur at different rates
an appropriate tracks are contain			I'm That them and destruction
2. Which is a macroscopic pro-			
a, colone		concentration.	
b. temperature	a.	pH	
3. A system reaches equilibrius	on. Which most be try	Catalogical Section 2	
a. it is open		it is either open or cle	
b, it is closed	4.	it is open and has pre-	PANIER:
Ice cobes are melting in a gl system?	ass of water with a lid	f on it. Why does this	s situation <u>not</u> represent an equilibriu
a. It is not closed system.			
 b. The equilibrium cannot be c. The temperature is not con 		opposite dispetion.	
d. The rates of opposing ch			
5. Which of these is NOT an er			
a. H ₂ O(1)eerH ₂ O(g)		$N_2O_2(g)$ \leftrightarrow $2NO_2(g)$	and the second s
b. $C(x) + CO_2(g) e + CO_2(g)$	-0.	C ₂ H ₂ OH(1)++C ₂ H ₂ OI	rs(g)
6. In the equilibrium constan	it expression below, w	had do the exponent	s represent?
(C) (D) a			
$K_{eq} = \frac{[C]^n [D]^n}{[A]^n [B]^n}$			
a, the concentrations of reac	tunts and products c.	the rates of the forwa	and and reverse reactions.
b. the equilibrium constant	-di	the stok blometric c	officients from the balanced equation
7. Which is tree of the pemeri-	cal value of the equilit	between constant, K., 7	T
a. It represents the relative	concentration of pro-	elects compared will	h reactants.
b. It represents the ratio of r			
c. It is measured in the SI us	sit, mole.		
d. Both a and c are true.			
8. What does it mean if $K_{**} < 1$	L at equilibrium?		
a. The concentration of re-		the concentration of	of products.
b. The concentration of prod	lacts is greater than the	concentration of reac	Carylin.
c. The rate of the forward re-			
st. The rate of the reverse rea	action in greater than th	e rate of the forward r	Propertional
9. When the equilibrium const the initial concentr		approximate concer	stration of reactant at equilibrium is
a. greater than		The same on	
b. less than	d.	exactly 100 times	
10. If Q _{at} > K _{at} , then			
a, the system is at equilibriu	_	of Ohio Station of I	products to reactants is greater than I
b. the ratio of products to re-			shifts low and product formation
11. Use the following informati			
The following reaction is an	example of homogene	sous equilibrium;	
$H_0(g) + I_0(g) \Leftrightarrow 2HI(g)$			
	The state of the s		
If the pressure is hadved, he a shift forward.		e and the temperatur not be affected	re remains constant, the equilibrium

12. Which condition is NOT a condition for an equilibrium system?

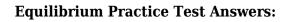
a. equal rates for opposing changes

b. a closed system

d. constant temperature

Equilibrium Practice Test Answers

CH Cherryholmes



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In some sort of inundated with monitors and the cacophony of instant connection, the profound power and emotional resonance of verbal art often disappear into obscurity, eclipsed by the continuous barrage of noise and distractions. Yet, nestled within the musical pages of **Equilibrium Practice Test Answers**, a charming function of literary beauty that pulses with natural thoughts, lies an unique journey waiting to be embarked upon. Composed by way of a virtuoso wordsmith, that magical opus manuals visitors on a mental odyssey, lightly revealing the latent potential and profound impact embedded within the delicate web of language. Within the heart-wrenching expanse with this evocative analysis, we shall embark upon an introspective exploration of the book is central subjects, dissect their captivating writing type, and immerse ourselves in the indelible impression it leaves upon the depths of readers souls.

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