

Skill – The student will be able to calculate the equilibrium constant, given the equilibrium concentrations which refer to the equation in level 2 and tell if the products or reactants are favored.

When a reaction is said to have reached equilibrium, the ratio of the concentration of products to reactants is equal to a value called the equilibrium constant (K). A small value of K indicates that the concentration of products is small and the concentration of reactants is large. We say the equilibrium lies to the left because there are more reactants at equilibrium. If the equilibrium constant is large, this indicates that there is a high concentration of products and a low concentration of reactants. In this situation we say the equilibrium lies to the right because there are more products at equilibrium.

Balance the equations and use the info below to calculate the equil. constant (K) for the following reactions. Circle whether the equilibrium would shift to the right or to the left.



Skill – The student will be able to use K_{sp} correctly in an equilibrium calculation.
Write the balanced chemical equations describing the dissolving of the following solids in water. Write the K_{sp} expression for each.



Use your knowledge of the relationship between K_{sp} , molarity, & solubility (g/L) to solve the following:



Equilibrium Unit Webquest And Notes Answers

Terry C. Jones



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Equilibrium Unit Webquest And Notes Answers Book Review: Unveiling the Magic of Language

In an electronic digital era where connections and knowledge reign supreme, the enchanting power of language has are more apparent than ever. Its power to stir emotions, provoke thought, and instigate transformation is actually remarkable. This extraordinary book, aptly titled "**Equilibrium Unit Webquest And Notes Answers**," written by a very acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound affect our existence. Throughout this critique, we will delve into the book is central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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