

Chapter 16: Equilibrium/Thermodynamics problems

Section 16.1 - 16.2

1. Write the equilibrium constant expression for K_c for the following reactions:



2. Write the equilibrium constant expression for K_p for $\text{H}_2\text{(g)} + \text{Cl}_2\text{(g)} \rightleftharpoons 2\text{HCl(g)}$

3. Write the chemical equation for ammonia formation from nitrogen and hydrogen. K_c is 0.50 at 400°C.



Calculate K_p for this reaction at the same temperature.

4. If at equilibrium $2\text{NO}_2\text{(g)} \rightleftharpoons \text{N}_2\text{(g)} + 2\text{O}_2\text{(g)}$ at a temperature of 230°C, the mass of oxygen at 0.050 mol O_2 . Calculate the value of K_c .

5. Write the equilibrium constant expressions for the acid dissociation of the following acids:



Equilibrium Ap Test Practice Problems Answers

Arturo Cuomo



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