

**Activity 4: Bonding and Polarity**  
**Data Table 2**

Bonding Pair	Electronegativity Difference	Type of Bond	Bond Polarity
H and H	0	Non - Polar Covalent	
B and F	2	Ionic	
Ge and F	2.2	Ionic	
C and H	0.4	Polar Covalent	$C \leftarrow H$
N and H	0.9	Polar Covalent	$N \leftarrow H$
H and O	1.4	Polar Covalent	$H \rightarrow O$
P and Cl	0.9	Polar Covalent	$P \rightarrow Cl$
S and F	1.5	Polar Covalent	$S \rightarrow F$
<del>Xe</del> and F	1.4	Polar Covalent	$Xe \rightarrow F$
O and O	0	Non - Polar Covalent	
C and O	1	Polar Covalent	$C \rightarrow O$
C and F	1.5	Polar Covalent	$C \rightarrow F$
Sn and Cl	1.2	<del>Gonic</del>	$Sn \rightarrow Cl$
Al and O	2	<del>Gonic</del>	
Na and Cl	2.1	<del>Gonic</del>	
Mg and O	2.3	<del>Gonic</del>	
Mg and Cl	1.8	<del>Gonic</del>	

# Electronegativity And Polarity Answers

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