# Practice Problems (Chapter 8): Electron Configuration

CHEM IA

1. Write complete electron configurations for the following atoms:

a) F 1s<sup>2</sup> 2s<sup>2</sup> 2p<sup>5</sup>

- b) A1
- c) V
- d) Cu
- e) Kr
- Write <u>noble gas</u> electron configurations for the following atoms:

3. How many core, outer, and valence electrons do each of the following atoms have? Core Outer Valence

- a) Cd [Kr] 5s<sup>2</sup> 4d<sup>10</sup>
  b) 1
  c) Ba
  d) Rn
- a) H 0 1 1
  b) Si 10 4 4
  c) Ti 18 2 4
  d) Br 28 7 7 (3d e are core e for rep, elements)

11

36

- 4. Write orbital notations for the following atoms:
  - a) O  $\frac{11}{1s}$   $\frac{11}{2s}$   $\frac{1}{2p}$   $\frac{1}{3s}$   $\frac{1}{3p}$   $\frac{1}{4s}$   $\frac{1}{3d}$

e) Ag

am exceeption):

- b) P  $\frac{1l}{ls} \frac{1l}{2s} \frac{1l}{2p} \frac{1l}{3s} \frac{1}{3p} \frac{1}{3p} \frac{1}{4s} = -\frac{1}{3d} = -\frac{1}{3d}$
- c) Cr  $\frac{1}{\text{Is}}$   $\frac{1}{2\text{s}}$   $\frac{1}{2\text{p}}$   $\frac{1}{3\text{s}}$   $\frac{1}{3\text{p}}$   $\frac{1}{4\text{s}}$   $\frac{1}{2}$   $\frac{1}$ 
  - d) Co  $\frac{11}{1s}$   $\frac{11}{2s}$   $\frac{11}{2p}$   $\frac{11}{3s}$   $\frac{11}{3p}$   $\frac{11}{4s}$   $\frac{11}{1}$   $\frac{11}{2}$   $\frac{1}{3d}$   $\frac{1}{2}$

# **Electron Configuration Problems**

**David Baud** 



# **Electron Configuration Problems** Book Review: Unveiling the Magic of Language

In a digital era where connections and knowledge reign supreme, the enchanting power of language has are more apparent than ever. Its ability to stir emotions, provoke thought, and instigate transformation is truly remarkable. This extraordinary book, aptly titled "**Electron Configuration Problems**," compiled by a highly acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound affect our existence. Throughout this critique, we shall delve to the book is central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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