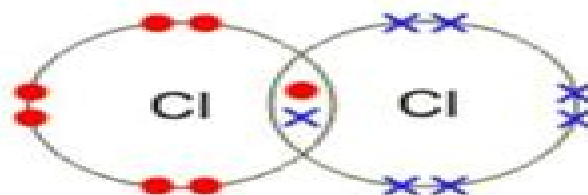


Covalent bonding

A non-metal metal and another non-metal can bond together by **sharing** electrons so that both electrons can achieve a **full outer shell**. This is called covalent bonding. To represent the atoms involved in the bonding we draw crosses for the electrons on one of the atoms and dots for the electrons on the other atoms as shown here.



Instructions

NB: It may be easiest to draw each atom out individually first to work out how many electrons will be shared

1. Draw the circles (outer shells) and overlap them (like a Venn diagram)
2. Write the symbols of the atoms involved in the middle of each circle
3. Draw the **electrons** for one atom as dots (o) and electrons for the other atom as crosses (x).
4. Any shared electrons are drawn in the **overlapping** section.
5. Count up the electrons for EACH atom and check that they each now have a **full outer shell** (2 electrons for the first shell and 8 for the next 2 shells)

TASK: Draw dot and cross diagrams for the following covalent compounds.

F ₂		CH ₄	
HCl		CO ₂	
H ₂ S		H ₂ O	
HF		NH ₃	

Challenge - O₂

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Sir Tarun Rupani



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Bonding Aligned to Next Generation Science Standards NGSS and other state standards *Metal-Metal Bonding* Gerard Parkin, 2010-03-04 None **New Concepts of Metallic Bonding** Mark Herbert McAdon, William A. Goddard, California Institute of Technology. Division of Chemistry and Chemical Engineering, 1988 **Metallic Bonds in Chemistry** Saeed Farrokhpay, 2023-12 This book serves as a comprehensive and invaluable guide for students researchers and professionals interested in understanding the fundamental principles of metallic bonding It explains the topic by presenting clear illustrations examples and case studies Metallic bonding is an important concept in chemistry and it forms the basis for understanding the structure properties and applications of metals in various industries from materials science and engineering to electronics and beyond It starts with a solid foundation by exploring the basic principles and theories that govern the bonding between metal atoms It also covers the relevant atomic structure and electronic configurations of metals to explain the factors affecting the metallic bonds formation In addition the crystal structures of the metals and their mechanical and thermal conduction properties are discussed Additionally the unique characteristics of metallic bonding in transition metals is covered due to their complex bonding patterns Finally the diverse applications of metallic bonding along with future directions in the field are fully discussed Metal-metal Bonding, 2010 John Berry Metal Metal Bonds in Chains of Three or More Metal Atoms From Homometallic to Heterometallic Chains Malcolm Chisholm Electronically Coupled MM Quadruple Bonded Complexes of Molybdenum and Tungsten Philip Power Transition Metal Complexes Stabilized by Bulky Terphenyl Ligands Applications to Metal Metal Bonded Compounds Gerard Parkin Metal Metal Bonding in Bridging Hydride and Alkyl Compounds Roland Fischer and Gernot Frenking Structure and Bonding of Metal Rich Coordination Compounds Containing Low Valent Ga I and Zn I Ligands Mike Hill Homocatenation of Metal and Metalloid Main Group Elements Constandinos A Tsipis Aromaticity Antiaromaticity in Bare and Ligand Stabilized Rings of Metal Atoms Alexander Boldyrev All Transition Metal Aromaticity and Antiaromaticity **METALLIC BOND** NARAYAN CHANGDER, 2024-04-05 Note Anyone can request the PDF version of this practice set workbook by emailing me at cbsenet4u gmail com You can also get full PDF books in quiz format on our youtube channel <https://www.youtube.com/smartquizz> I will send you a PDF version of this workbook This book has been designed for candidates preparing for various competitive examinations It contains many objective questions specifically designed for different exams Answer keys are provided at the end of each page It will undoubtedly serve as the best preparation material for aspirants This book is an engaging quiz eBook for all and offers something for everyone This book will satisfy the curiosity of most students while also challenging their trivia skills and introducing them to new information Use this invaluable book to test your subject matter expertise Multiple choice exams are a common assessment method that all prospective candidates must be familiar with in today's academic environment Although the majority of students are accustomed to this MCQ format many are not well versed in it To achieve success in MCQ tests quizzes and trivia challenges one requires test taking techniques and skills in addition to subject knowledge It also

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