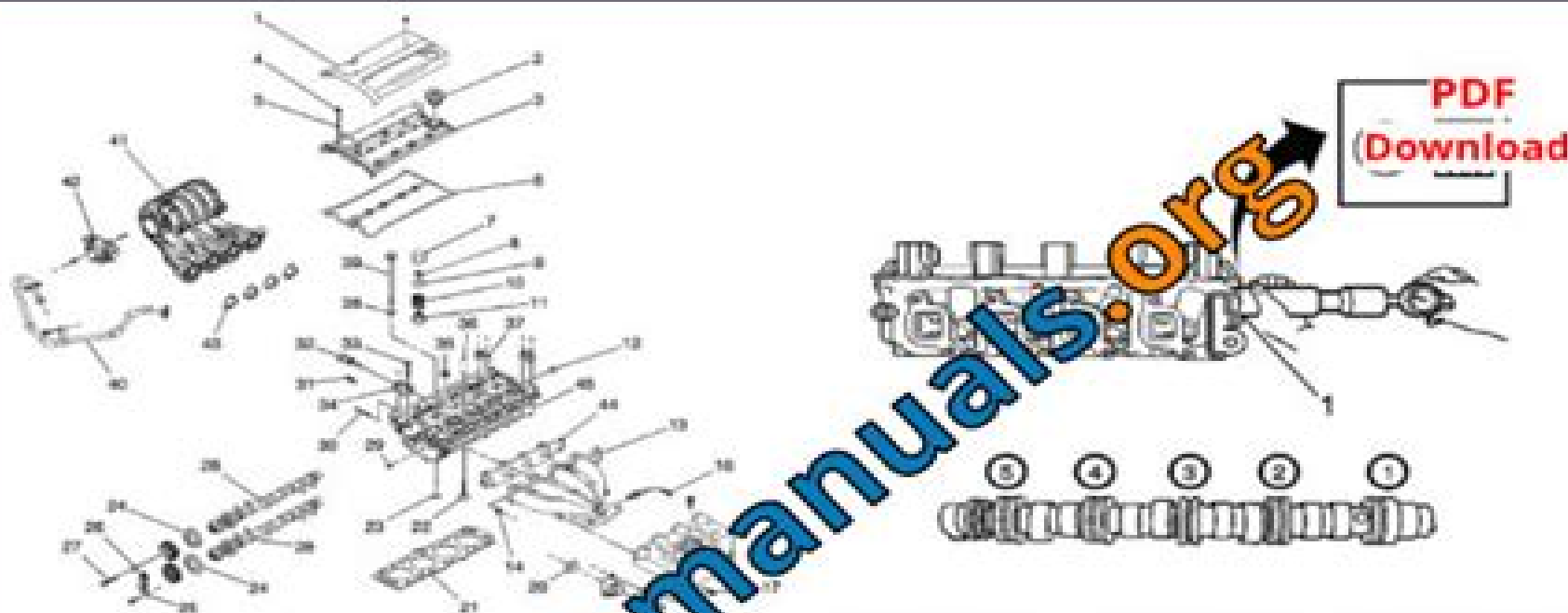


Daewoo Kalos 2002-2006 Workshop Manual



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Fig. 1: Upper End
Courtesy of GENERAL MOTORS CORP.

Callouts For Fig. 1

Callout	Component Name
1	Engine Cover
2	Oil Cap
3	Valve Cover
4	Valve Cover Bolt
5	Valve Cover Washer
6	Valve Cover Gasket
7	Hydraulic Valve Lash Adjuster
8	Valve Key
9	Valve Spring Retainer
10	Valve Spring
11	Valve Stem Oil Seal

Fig. 1B1: Measuring Inner Diameter Of Cylinder Head Journal With Bore Gage
Courtesy of GENERAL MOTORS CORP.

Using a bore gage (1), measure the inner diameter of the cylinder head journal at the 5 places. The clearance of the journal is measured by the difference between the outer diameter of the camshaft the inner diameter of the cylinder head journal part.

Specifications

- The journal clearance standard is 0.05-0.091 mm (0.0020-0.0036 in).
- The journal clearance limit is 0.15 mm (0.0059 in).
- The camshaft journal outer diameter (1) is 43.425-43.450 mm (1.7096-1.7106 in).
- The inner diameter cylinder head journal part (1) is 43.500-43.516 mm (1.7126-1.7132 in).
- The camshaft journal outer diameter (2) is 43.625-43.650 mm (1.7175-1.7185 in).

Kalos Engine Manual

TD Snyder



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 what do the following terms mean oxidation oxidation is a loss of electrons reduction reduction is a gain of electrons what is
 a helpful way to remember this

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web write the oxidation and reduction reactions for each redox reaction the first one is done for you 23 $\text{Fe}^{2+} + \text{Co} \rightarrow \text{Co}^{2+} + \text{Fe}$
 oxidation $\text{Co} \rightarrow \text{Co}^{2+} + 2\text{e}^-$ reduction $\text{Fe}^{2+} + 2\text{e}^- \rightarrow \text{Fe}$ 24 3 $\text{Ag} + \text{Ni} \rightarrow \text{Ni}^{3+} + 3\text{Ag}$ oxidation reduction 25 $\text{Cu}^{2+} + \text{Pb} \rightarrow \text{Pb}^{2+} + \text{Cu}$ oxidation reduction 26 O_2

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 magnitude of change then identify the element oxidized and reduced as well as the oxidizing agent and reducing agent a $\text{I}_2 + 5\text{O}_2 \rightarrow 5\text{CO}_2$
 $\text{I}_2 + 5\text{CO} \rightarrow 5\text{CO}_2 + \text{I}_2$

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 reduction reactions or redox reactions in this unit you will examine the oxidation reduction process and use the oxidation
 state and oxidation number concepts to identify redox reactions and to keep track of electrons

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 reducing agent also $2\text{Sr} + \text{O}_2 \rightarrow 2\text{SrO}$ Sr^0 to Sr^{2+} oxidized reducing agent O_2 to O^{2-} reduced ox ag $2\text{Li} + \text{S} \rightarrow \text{Li}_2\text{S}$ Li^0 to Li^{+1} oxidized red ag
 S^0 to S^{2-} reduced ox ag $3\text{Cs} + \text{Br}_2$