

Answer Sheet for Sample Problems for Chemistry Level I Final Exam 2013-2014 Study Guide

Electrons in Atoms – Chapter 5

1. What is the frequency of green light, which has a wavelength of $4.90 \times 10^{-7} \text{ m}$?

$$c = \lambda \nu \quad \lambda = 4.90 \times 10^{-7} \text{ m}, c = 3.00 \times 10^8 \text{ m/s} \quad \nu = ? \text{ Hz} \quad \nu = \frac{c}{\lambda} = \frac{3.00 \times 10^8 \text{ m/s}}{4.90 \times 10^{-7} \text{ m}}$$

$$\underline{6.12 \times 10^{14} \text{ Hz}}$$

2. An x-ray has a wavelength of $1.15 \times 10^{-10} \text{ m}$. What is its frequency?

$$c = \lambda \nu \quad \lambda = 1.15 \times 10^{-10} \text{ m}, c = 3.00 \times 10^8 \text{ m/s} \quad \nu = ? \text{ Hz} \quad \nu = \frac{c}{\lambda} = \frac{3.00 \times 10^8 \text{ m/s}}{1.15 \times 10^{-10} \text{ m}}$$

$$\underline{2.61 \times 10^{18} \text{ Hz}}$$

3. Which type of electromagnetic radiation has the most energy: Microwaves, X-Rays, or Visible light? Why?

Ans: X-Rays, b/c they have the shortest wavelength and highest frequency of the types listed and frequency and energy are directly related.

4. What is the energy of a $9.50 \times 10^{13} \text{ Hz}$ radiation?

$$E = h\nu \quad \nu = 9.50 \times 10^{13} \text{ Hz}, h = 6.626 \times 10^{-34} \text{ J}\cdot\text{s} \quad E = ? \text{ J}$$

$$E = h\nu = (6.626 \times 10^{-34} \text{ J}\cdot\text{s}) (9.50 \times 10^{13} \text{ s}^{-1}) = \underline{6.29 \times 10^{-19} \text{ J}}$$

5. The element with the electron configuration of $[\text{Kr}]5s^2 4d^{10} 5p^1$ is indium (In)

6. Barium (Ba)

7. Write out the electron configurations for the following elements: (For the exam make sure you know how to do both the complete and the noble gas short cut for electron configurations)

a. Tc **Ans: $1s^2 2s^2 2p^6 3s^2 3p^4 4s^2 3d^5 4p^6 5s^2 4d^5$ or $[\text{Kr}] 5s^2 4d^5$**

b. Aluminum **Ans: $1s^2 2s^2 2p^6 3s^2 3p^1$ or $[\text{Ne}] 3s^2 3p^1$**

8. Why are Cr and Cu exceptions to writing the electron configurations of elements?

They are exceptions because instead of following the pattern of filling up the d orbitals Cr actually has a half filled s and d orbitals and Cu has a half filled s and completely filled d orbitals. The reason why this occurs is because the electrons in the d-sublevel are more stable with this electron configuration than the expected configuration.

9. a. Ca

b. As

c. Sn

a. 2 Ca.

b. 5 .As.

c. 4 . Sn .

Periodic Table and Periodic Trends – Chapter 6

1. Compare & contrast how Mendeleev's periodic table was organized to how Moseley organized it.

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Regents Exams and Answers Physics Physical Setting Revised Edition Barron's Educational Series, Miriam Lazar, 2021-01-05 Barron's Regents Exams and Answers Physics provides essential review for students taking the Physics Regents including actual exams administered for the course thorough answer explanations and comprehensive review of all topics This edition features Eight actual administered Regents exams so students can get familiar with the test Comprehensive review questions grouped by topic to help refresh skills learned in class Thorough explanations for all answers Score analysis charts to help identify strengths and weaknesses Study tips and test taking strategies *PGT Commerce - Unit 6 : Finance & Banking (2025 Edition) - Self Study Series Book* Dheeraj Kumar Singh, The book is specially written for PGT Commerce examinations like DSSSB NVS HTET Level III HPSC and All other state level teaching examinations The book is an attempt to clarify the theoretical concept and provide practical problem solving aptitude to crack the objective type examinations There is a chapter wise segregation of the exam questions **Regulatory Foundations for the Food Protection Professional** Julia Bradsher, Gerald Wojtala, Craig Kaml, Christopher Weiss, David Read, 2015-07-13 Regulatory Foundations for the Food Protection Professional is a comprehensive guide for the entry level food protection professional FPP working in either the public or private sector The book can also serve as a foundation for students in academic programs preparing for a career in food protection Additionally as the Food Safety Modernization Act FSMA is implemented this book will provide valuable information for countries wishing to export foods and food ingredients to the U S and comply with U S food safety regulations The book is based on the Entry Level component of the National Curriculum Framework for regulators created by the International Food Protection Training Institute IFPTI located in Battle Creek Michigan The Entry Level component of the National Curriculum Framework contains more than twenty content areas including Epidemiology Microbiology Labeling Food Defense Awareness Program Standards Environmental Health and Safety Sampling and Allergens Each chapter is divided into specific learning objectives aimed at equipping the entry level FPP with the knowledge and skills necessary to successfully perform his or her job whether in the public or private sector and whether in food safety or food defense Established in 2009 IFPTI is improving public health by building competency based training and certification systems and cultivating leadership for the food protection community worldwide Our mission is to enhance public health by improving the protection of the world's food supply through training certification thought leadership and technology See more at www.ifpti.org **Regents Physics--Physical Setting Power Pack Revised Edition** Barron's Educational Series, Miriam A. Lazar, Albert Tarendash, 2021-01-05 Barron's Regents Physics Power Pack provides comprehensive review actual administered exams and practice questions to help students prepare for the Physics Regents exam This edition includes Two actual Regents exams online Regents Exams and Answers Physics Physical Setting Four actual administered Regents exams so students have the practice they need to prepare for the test Review questions

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scholars in both the neurosciences and organizational sciences who have adopted various approaches to study the cognitive mechanisms mediating the social behavior that we see within organizations. Such an approach has been termed by ourselves and others as organisational cognitive neuroscience. In recent years there has been a veritable increase in studies that have explored the cognitive mechanisms driving such behaviors and much progress has been made in understanding the neural underpinnings of processes such as financial exchange, risk awareness and even leadership. However, while these studies are informative and add to our understanding of human cognition, they fall short of providing evidence-based recommendations for practice. Specifically, we address the broader issue of how the neuroscientific study of such core social behaviors can be used to improve the very way that we work. To address these gaps in our understanding, the chapters in this book serve as a platform that allows scholars in both the neurosciences and the organizational sciences to highlight the work that spans across these two fields. The consolidation of these two fields also serves to highlight the utility of a singular organizational cognitive neuroscience. This is a fundamentally important outcome of the book as the application of neuroscience to address economically relevant behaviors has seen a variety of fields evolve in their own right, such as neuromarketing, neuroeconomics and so forth. The use of neuroscientific technologies, in particular fMRI, has indeed led to a bewildering and somewhat suffocating proliferation of new approaches; however, the speed of such developments demands that we must proceed carefully with such ventures or risk some fundamental mistakes. The book that you now hold will consolidate these new neuroscience-based approaches and, in doing so, highlight the importance of this approach in helping us to understand human social behavior in general. Taken together, the chapters provide a framework for scholars within the neurosciences who wish to explore the further opportunities that the study of organisational behavior may provide.

Marketing Theory Michael J Baker, Michael Saren, 2016-05-16. Marketing Theory introduces and explains the role of theory in marketing by uncovering its historical, disciplinary underpinnings, subfields, discourses and debates. From strategy and ethics to digital marketing and consumer behaviour, leading marketing experts shine a light on what can be a challenging perspective of marketing. In this new Third Edition, there are up-to-date examples from global companies such as Pepsi, Amazon and H&M. Entirely new chapters on Digital and Social Media Marketing and Service-Dominant Logic (SD-L) and contributions from Global Specialists including Bob Lusch, Patrick Murphy and Susan Hart. Ideal for upper-level undergraduate and postgraduate marketing students studying marketing theory, critical marketing and the history of marketing modules.

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