

(e) When reacted with sulfuric acid, lactic acid, C<sub>3</sub>H<sub>6</sub>O<sub>3</sub>, loses water to form a compound with the molecular formula C<sub>6</sub>H<sub>8</sub>O<sub>4</sub>. The infrared spectrum of this compound shows an absorption at 1735–1750 cm<sup>-1</sup> and no absorptions above 3000 cm<sup>-1</sup>.

Suggest a structure for this compound, giving your reasoning.

- (f) When malic acid is heated, it loses water to form an acid anhydride with the molecular formula C<sub>4</sub>H<sub>2</sub>O<sub>3</sub>.
  - Suggest a structure for this compound.

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