

Answers

$Q = mc\Delta T$, where Q = heat energy, m = mass, and ΔT = change in temp.
Remember, $\Delta T = (T_{\text{final}} - T_{\text{initial}})$. Show all work and proper units.

1. A 15.75-g piece of iron absorbs 1086.75 joules of heat energy, and its temperature changes from 25°C to 175°C. Calculate the specific heat capacity of iron.

$$C = \frac{Q}{m(T_f - T_i)} = \frac{1086.75}{15.75(175-25)} = 0.46 \text{ J/g}^\circ\text{C}$$

2. How many joules of heat are needed to raise the temperature of 10.0 g of aluminum from 22°C to 55°C, if the specific heat of aluminum is 0.90 J/g°C?

$$Q = mC(T_f - T_i) = 10.0\text{g} (0.90\text{J/g}^\circ\text{C})(55-22) = 297 \text{ J}$$

3. Calculate the specific heat capacity of a piece of wood if 1500.0 g of the wood absorbs 67,500 joules of heat, and its temperature changes from 32°C to 57°C.

$$C = \frac{Q}{m(T_f - T_i)} = \frac{67500 \text{ J}}{(1500 \text{ g})(57-32)} = 1.8 \text{ J/g}^\circ\text{C}$$

4. 100.0 g of 4.0°C water is heated until its temperature is 37°C. Calculate the amount of heat energy needed to cause this rise in temperature.

$$Q = mC(T_f - T_i) = 100\text{g}(4.184\text{J/g}^\circ\text{C})(37 - 4) = 14000 \text{ J}$$

5. 25.0 g of mercury is heated from 25°C to 155°C, and absorbs 455 joules of heat in the process. Calculate the specific heat capacity of mercury.

$$C = \frac{Q}{m(T_f - T_i)} = \frac{455 \text{ J}}{(25\text{g})(155-25)} = 0.14 \text{ J/g}^\circ\text{C}$$



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Sh. Ye Mikeladze



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Numerical Solution of the Heat Equation by the Method of Heat Potentials Daniel Gordon Willis,1986 **Numerical Solutions and Applications of the Heat Equation** Anna Morgante,Meredith College (Raleigh, N.C.). Honors Program,2009 *Tables for Solution of the Heat-conduction Equation with a Time-dependent Heating Rate* A. E. Bergles,1962 *Calculation of Equation for Specific Heat and Specific Heat Ratio as a Function of Temperature Range Limit 600°R to 1800°R* ,1964 Heat Loss Calculation Guide Hydronics Institute,1976 A Numerical Solution of the Heat Transfer Equation Sh. Ye Mikeladze,1963 **Heat Loss Calculation Guide** International Telephone and Telegraph Corporation. Fluid Handling Division,1967 **Heat Loss Calculation Guide** Hydronics Institute,1998 *Heat Energy Calculation* ,2006 *Heat Loss Calculation Guide* Hydronics Institute,1987 **Solution of the Radiative Heat Transfer Equation by the Discrete Ordinates Method in Body-fitted Coordinates** Rahul Shah,1996 **Deviation trends between heat loss calculation methods** Ian Jackson,University of Manchester Institute of Science and Technology. Department of Building Engineering,1998

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