

PROBLEM 2.25^o

KNOWN: A known amount of gas undergoes a constant-pressure process in a piston-cylinder assembly beginning at a specified specific volume. The work is known.

Find: Determine the final volume.

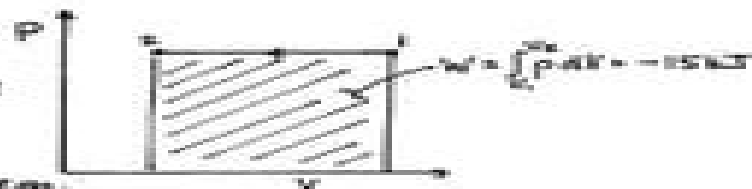
SCHEMATIC & GIVEN DATA:



$$v_1 = 0.2 \text{ m}^3/\text{kg}$$

$$W = -15 \text{ kJ}$$

$$P = 5 \text{ bar}$$



①

ASSUMPTIONS: (1) The gas is a closed system.
(2) Pressure is constant during the process.

ANALYSIS: Using Eq. 2.17

$$W = \int_{V_1}^{V_2} P dV = P(V_2 - V_1)$$

$$= P(V_2 - m v_1)$$

Solving for V_2 and inserting values

$$V_2 = \frac{W}{P} + m v_1$$

$$= \frac{(-15 \text{ kJ})}{(5 \text{ bar})} \left| \frac{1 \text{ bar}}{10^5 \text{ N/m}^2} \right| \left| \frac{10^3 \text{ N}\cdot\text{m}}{1 \text{ kJ}} \right| + (0.25 \text{ kg})(0.2 \text{ m}^3/\text{kg})$$

$$= 0.02 \text{ m}^3 \quad V_2$$

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Jack Winnick



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