

Chemistry 11

Atomic Theory

The Rutherford Model

To account for Rutherford's discovery of the nucleus, the Rutherford model was developed. This model shows a small, positively charged ball at the center of the atom, with most of the mass concentrated there.

Notes

Activity 1.1: Rutherford's Model

Activity 1.1: Rutherford's Model
This activity is designed to help you understand the Rutherford model of the atom.



Model 1: A simple model of an atom with a central nucleus and three electrons. This model is based on the Rutherford model of the atom.

Model 2: A more complex model of an atom with a central nucleus and three electrons. This model is based on the Rutherford model of the atom.

Objectives

After completing this activity, you should be able to:

- 1. Describe the Rutherford model of the atom.
- 2. Explain the structure of an atom.



After completing this activity, you should be able to:

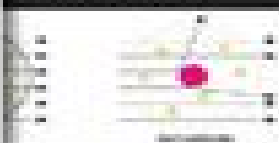
1. Describe the Rutherford model of the atom.

2. Explain the structure of an atom.



Preparation

Materials: A simple model of an atom with a central nucleus and three electrons.



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