

ACS Biochemistry Final

pKa (chapter 3)

measure of the tendency of a group to give up a proton (acidity); tendency decreases tenfold as pKa increases by one unit

thioester bond (1)

compounds with functional group C-S-CoA-C (eg, in acetyl-CoA)

zwitterion (3)

dipolar ion with both positive and negative groups but overall neutral charge; can act as either an acid or base

Isoelectric point (pI) (chapter 3)

the characteristic pH at which the net electric charge is zero

SDS - Sodium Dodecyl Sulfate (chap 3)

detergent used to unfold proteins and give them uniform negative charge

SDS Page (3)

chromatography used to separate proteins based on mass. light proteins travel fast than heavier ones

PCR - Polymerase chain reaction (3)

copies DNA multiple times to increase sample size

Isoelectric focusing (chapt 3)

procedure used to determine the isoelectric pt (pI) of a protein. Protein migrates through gel until pH = pI (net charge = 0)

Two-Dimensional Electrophoresis (3)

combines isoelectric focusing and SDS electrophoresis; separates proteins by both molecular weight and pI

specific activity (3)

number of enzyme units per mg of total protein (a measure of enzyme purity)

activity (3)

total units of a certain enzyme in a solution

Peptide bonds (chapter 4)

C-N bond with double bond character due to resonance (C-N bond cannot rotate, and is planar)

Edman degradation (3)

used in the sequencing of polypeptides; labels and removes ONLY the amino-residue from a polypeptide. carried out in a machine called a sequenator

ϕ in peptide bonding (chapter 4)

angle around the α -carbon - amide nitrogen bond

ψ in peptide bonding (chapter 4)

angle around the α -carbon - carbonyl carbon bond

Ramachandran Plot (4)

shows favorable ϕ - ψ angle combinations. 3 main "wells" for α -helices, β -sheets, and left handed α -helices

Levinthal's Paradox (4)

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1. Understanding the eBook How To Study For Acs Biochem Final And
 - The Rise of Digital Reading How To Study For Acs Biochem Final And
 - Advantages of eBooks Over Traditional Books
2. Identifying How To Study For Acs Biochem Final And
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an How To Study For Acs Biochem Final And
 - User-Friendly Interface
4. Exploring eBook Recommendations from How To Study For Acs Biochem Final And
 - Personalized Recommendations
 - How To Study For Acs Biochem Final And User Reviews and Ratings
 - How To Study For Acs Biochem Final And and Bestseller Lists
5. Accessing How To Study For Acs Biochem Final And Free and Paid eBooks
 - How To Study For Acs Biochem Final And Public Domain eBooks
 - How To Study For Acs Biochem Final And eBook Subscription Services
 - How To Study For Acs Biochem Final And Budget-Friendly Options
6. Navigating How To Study For Acs Biochem Final And eBook Formats

- ePub, PDF, MOBI, and More
 - How To Study For Acs Biochem Final And Compatibility with Devices
 - How To Study For Acs Biochem Final And Enhanced eBook Features
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 - Highlighting and Note-Taking How To Study For Acs Biochem Final And
 - Interactive Elements How To Study For Acs Biochem Final And
 8. Staying Engaged with How To Study For Acs Biochem Final And
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers How To Study For Acs Biochem Final And
 9. Balancing eBooks and Physical Books How To Study For Acs Biochem Final And
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection How To Study For Acs Biochem Final And
 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
 11. Cultivating a Reading Routine How To Study For Acs Biochem Final And
 - Setting Reading Goals How To Study For Acs Biochem Final And
 - Carving Out Dedicated Reading Time
 12. Sourcing Reliable Information of How To Study For Acs Biochem Final And
 - Fact-Checking eBook Content of How To Study For Acs Biochem Final And
 - Distinguishing Credible Sources
 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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