
ACS Gen Chem 1 Exam

Ion - A species of an element in which the number of electrons does not equal the number of protons.

Isotope - Each isotope of an atom has a different number of neutrons.

Isoelectric - Two atoms with the same charge.

Isotopic - Two atoms with the same number of neutrons.

Transition Metal Orbitals - The 4s orbital is filled before the 3d orbital, but listed after.

Paramagnetic - Atoms that possess a permanent magnetic charge due to the presence of unpaired electrons.

Electropositive - Tending to lose electrons and form positive ions.

Tetrahedral Bond Angle - 109.5°

Resonance Structures - Two equally correct arrangements of one molecule's electrons; generally the molecule exists as an average of the two.

Polar Molecule - A molecule that is mostly positive on one side and mostly negative on the other.

sp^3 Orbital - An orbital composed of one s orbital and three p orbitals; can hold eight total electrons. Tetrahedral.

Trigonal Planar - The way that three objects will arrange themselves around a central atom. Non-polar.

Free Radicals - Compounds with an odd number of electrons in their Lewis Structure. Eg. NO_2

Incomplete Octets - Occur in elements that can form stable bonds with less than eight electrons. Eg. BF_3

Expanded Octets - Occur in the third row of the periodic table and beyond in elements that can form stable bonds with more than eight electrons. Eg. XeF_2

Hydrogen Bonding - When hydrogen is attracted to electronegative atoms.

General Chem 1 Exam Study Guide

G Thomas



General Chem 1 Exam Study Guide:

Unveiling the Power of Verbal Artistry: An Mental Sojourn through **General Chem 1 Exam Study Guide**

In a world inundated with displays and the cacophony of instant conversation, the profound energy and emotional resonance of verbal artistry frequently fade into obscurity, eclipsed by the constant onslaught of noise and distractions. However, nestled within the lyrical pages of **General Chem 1 Exam Study Guide**, a interesting perform of literary brilliance that impulses with raw emotions, lies an remarkable trip waiting to be embarked upon. Composed by way of a virtuoso wordsmith, this interesting opus manuals readers on a mental odyssey, lightly exposing the latent possible and profound impact embedded within the intricate web of language. Within the heart-wrenching expanse with this evocative evaluation, we can embark upon an introspective exploration of the book is main subjects, dissect their captivating writing model, and immerse ourselves in the indelible impression it leaves upon the depths of readers souls.

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