



Student Exploration: Solubility and Temperature

Background: Temperature affects chemical, biological, and physical processes. Solubility, the ability of a substance to dissolve in a solvent, is one such process.

Prior Knowledge: Questions about how temperature affects solubility.

1. When sugar is added to water, it dissolves. If you add more sugar, it will dissolve until it reaches a point where no more sugar can dissolve. This is called a **saturated solution**. If you add more sugar to a saturated solution, it will not dissolve. This is called a **supersaturated solution**.
If you heat a saturated solution, more sugar can dissolve. If you cool it, the extra sugar will precipitate out. This is called **precipitation**.
2. When sugar is added to water, it dissolves. If you add more sugar, it will dissolve until it reaches a point where no more sugar can dissolve. This is called a **saturated solution**. If you add more sugar to a saturated solution, it will not dissolve. This is called a **supersaturated solution**.
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If you heat a saturated solution, more sugar can dissolve. If you cool it, the extra sugar will precipitate out. This is called **precipitation**.

Warm-up

At room temperature, sugar is **soluble** in water. If you heat the water, the sugar will dissolve faster. If you cool the water, the sugar will precipitate out. This is called **precipitation**. The process of a substance dissolving in a solvent is called **dissolution**. The process of a substance coming out of a solution is called **precipitation**.

Temperature affects the rate of dissolution. If you heat a solution, the rate of dissolution increases. If you cool a solution, the rate of dissolution decreases.



1. In this solution, which is the solute? _____ (The substance that is dissolved.)
Which is the solvent? _____ (The substance that does the dissolving.)
2. When you heat a solution, the rate of dissolution increases. If you cool a solution, the rate of dissolution decreases. This is called **precipitation**.
If you heat a saturated solution, more solute can dissolve. If you cool it, the extra solute will precipitate out. This is called **precipitation**.

Explore Learning Solubility And Temperature Gizmo Answers

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Explore Learning Solubility And Temperature Gizmo Answers:

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