Naming Review Practice

Directions: On the left of each question, identify if the compound/molecule is lonic (write an I) or Covalent (write a C). Then, either write the chemical name of the molecule (if given the formula) or the chemical formula of the molecule (if given the name)

Covalent	1. CS ₃	Carbon Trisulfide	
Ionic	2. TiO2	Titanium IV Oxide	
Ionic	3. CaCO ₃	Calcium Carbonate	
Ionic	4. Cu(NO ₂) ₂	Copper II Nitrite	
Ionic	5. Sn(CN)4	Tin IV Cyanide	
Ionic	6. Ag ₃ SO ₃	Silver Sulfite	
Ionic	7. NiSe ₂	Nickel IV Selenide	
Covalent	8. P ₃ O ₅	Triphosphorus Pentoxide	
Ionic	9. Sn ₃ (PO ₄) ₂	Tin II Phosphate	
Covalent	10. N ₇ Cl ₃	Heptanitrogen Trichloride	
Ionic	11. Manganese (III) bromsde		MnBr ₃
Covalent	12. Sulfur dioxide		SO ₂
Ionic	13. Tin (IV) sulfate		Sn(SO ₄) ₂
Ionic	14. Zinc hydroxide		Zn(OH) ₂
Covalent	15. Carbon tetrachloride		CCL ₄
Ionic	16. Barium cyanide		Ba(CN) ₂
Ionic	17. Cobalt (VI) Chromate		Co(CrO ₄) ₃
Ionic	18. Lithium nitrite		LiNO ₂
Covalent	19. Trisulfur hexafluoride		S ₃ F ₆
Ionic	20. Aluminum oxalate		$Al_2(C_2O_4)_3$

Naming Review Guide Answer Key

Elizabeth C. Axford

Naming Review Guide Answer Key:

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