










Class	General Formula	Example	Common Name (Systematic Name)	Common Suffix/Prefix (Systematic)
<b>Hydrocarbons</b>				
Alkanes	$RH$	$CH_3CH_3$	ethane	-ane
Alkenes	$RR'C=CR''R'''$	$H_2C=CH_2$	ethylene (ethene)	-ene
Alkynes	$RC\equiv CR'$	$HC\equiv CH$	acetylene (ethyne)	(-yne)
Arenes	$ArH^a$		benzene	-ene
<b>Halogen-Containing Compounds</b>				
Alkyl halides	$RX$	$CH_3CH_2Cl$	ethyl chloride (chloroethane)	halide (halo-)
Aryl halides	$ArX^a$		chlorobenzene	halo-
<b>Oxygen-Containing Compounds</b>				
Alcohols	$ROH^a$	$CH_3CH_2OH$	ethyl alcohol (ethanol)	-ol
Phenols	$ArOH^a$		phenol	-ol
Ethers	$ROR'$	$H_3CH_2COCH_2CH_3$	diethyl ether	ether
Aldehydes	$RCHO$		acetaldehyde (ethanal)	-aldehyde (-al)
Ketones	$RR'C=O$		acetone (2-propanone)	-one
Carboxylic acids	$RCO_2H$		acetic acid (ethanoic acid)	-ic acid (-oic acid)
<b>Carboxylic Acid Derivatives</b>				
Esters	$RCO_2R'$		methyl acetate (methyl ethanoate)	-ate (-oate)
Amides	$RCONHR'$		N-methylacetamide	-amide
<b>Nitrogen-Containing Compounds</b>				
Amines	$RNH_2, RNHR', RNR'R''$	$CH_3CH_2NH_2$	ethylamine	-amine
Nitriles	$RC\equiv N$	$H_3CC\equiv N$	acetonitrile	-nitrile
Nitro compounds	$ArNO_2^a$		nitrobenzene	nitro-

<sup>a</sup>R indicates an alkyl group <sup>a</sup>Ar indicates an aryl group.

# Naming Compounds Cheat Sheet

**A Gutmann**



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