

Three Hydrogen-like Atoms – Simultaneous Quantities

Electromagnetic Radiation

Consider now the hydrogen atom, which emits energy levels, light, and transitions and also is the following quantities in terms of its understanding:

Quantum is a concept that describes the behavior of particles, such as the photon, which is the quantum of electromagnetic radiation. The photon is a particle that carries energy and momentum, and it is the smallest unit of electromagnetic radiation that can be emitted or absorbed by an atom.

Quantity	Hydrogen Atom	Hydrogen Atom
Energy level	Quantum of electromagnetic radiation	Quantum of electromagnetic radiation
Transition	Quantum of electromagnetic radiation	Quantum of electromagnetic radiation
Photon	Quantum of electromagnetic radiation	Quantum of electromagnetic radiation
Electromagnetic radiation in discrete units	Quantum of electromagnetic radiation	Quantum of electromagnetic radiation
Energy of the photon or level	$E = \frac{hc}{\lambda}$ or $E = \frac{hc}{\lambda_0}$	$E = \frac{hc}{\lambda}$ or $E = \frac{hc}{\lambda_0}$

Quantum is a concept that describes the behavior of particles, such as the photon, which is the quantum of electromagnetic radiation. The photon is a particle that carries energy and momentum, and it is the smallest unit of electromagnetic radiation that can be emitted or absorbed by an atom. The photon is a particle that carries energy and momentum, and it is the smallest unit of electromagnetic radiation that can be emitted or absorbed by an atom.

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Jiyuan Zhang



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