(c) 350 cm, convergent
(d) 150 cm, divergent

16. Three resistances A, B and C each of 5Ω are connected to a battery of 10 V as shown in the figure. Calculate the current through C:

(a) 2 amp (b) 1 amp
(c) 3 amp (d) 0.6 amp

17. The device used to generate electrical energy is:
(a) Electric motor (b) Generator
(c) Galvanometer (d) Voltmeter

18. The number of molecules in 11 g of CO_2 are:
(a) 0.25×10^{23} (b) 0.50×10^{23}
(c) 1.00×10^{23} (d) 1.51×10^{23}

19. The value of charge / mass ratio of electron was determined by:
(a) W.K Roentgen (b) J.J.Thomson
(c) Marie Curie (d) Niels Bohr

20. Which of the following reactions is an example of combination reaction?
(a) $Zn(s) + CuSO_4(aq) \rightarrow ZnSO_4(aq) + Cu(s)$
(b) $BaCl_2(aq) + Na_2SO_4(aq) \rightarrow BaSO_4(s) + 2NaCl(aq)$
(c) $CaCO_3(s) \rightarrow CaO(s) + CO_2(g)$
(d) $C(s) + O_2(g) \rightarrow CO_2(g)$

21. The element 'X' and 'Y' have atomic numbers 12 and 17 respectively. Element 'X' reacts with element 'Y' to form compound with molecular formula:
(a) XY (b) XY_2
(c) XY_3 (d) X_2Y_3

22. The atomic radius (pm) of Li, Na, K and Rb varies in the order:
(a) Na<K<Rb<Li (b) K<Na<Li<Rb
(c) Li<Na<K<Rb (d) Rb<K<Na<Li

23. The electron affinity value (kJ mol^{-1}) of fluorine (F) is less than:
(a) Hydrogen (H) (b) Lithium (Li)
(c) Oxygen (O) (d) Chlorine (Cl)

24. Which of the following is an example of strong electrolyte?
(a) H_2CO_3 (b) NH_4OH
(c) NaCl (d) COOH – COOH

25. If pH of a solution changes from 5 to 4, the change in hydrogen ion concentration shall be:
(a) two times (b) five times
(c) ten times (d) twenty times

26. Bleaching powder is manufactured by the reaction of:
(a) $CaCl_2$ and $CaCO_3$ (b) Cl_2 and $Ca(OH)_2$
(c) Cl_2 and $CaSO_4$ (d) Cl_2 and $Ca(HCO_3)_2$

27. The hard glass is obtained by fusing:
(a) soda ash, sand and limestone
(b) a mixture of sand, lime, borax and alkali carbonates
(c) potassium carbonate and limestone
(d) potassium carbonate, lead oxide and sand

28. Which of the following metals is most reactive?
(a) Aluminium (b) lead
(c) Mercury (d) Silver

29. Which of the following alloys contains chromium?
(a) Steel (b) Stainless steel
(c) Magnalium (d) Brass

30. Which of the following is monomer of natural rubber?
(a) Chloroethene (b) Chloroprene
(c) Isoprene (d) Buta-1, 3-diene

31. Alkaline $KMnO_4$ oxidises propanone to:
(a) propanoic acid (b) ethanoic acid
(c) methanoic acid (d) oxalic acid

32. Which of the following metals can displace Zn from $ZnSO_4$ solution?
(a) Calcium (b) Copper
(c) Iron (d) Mercury

33. Heating of sodium ethanoate with soda lime yields:
(a) ethane (b) methane
(c) ethanol (d) methanol

34. The compound formed by the reaction of ethyne with bromine is:
(a) Br-CH=CH-Br (b) $Br-CH_2-CHBr_2$
(c) $CH_2=CH-Br$ (d) $Br_2CH-CHBr_2$

35. Cell were first discovered by Robert Hooke in the year:
(a) 1665 (b) 1674 (c) 1831 (d) 1839

36. The plant cells have a rigid cell wall that lies:
(a) Outside the plasma membrane
(b) Inside the plasma membrane
(c) In between the plasma membranes
(d) None of the above

37. During mitosis nucleolus and nuclear membrane are lost in stage:
(a) Prophase (b) Metaphase
(c) Anaphase (d) Telophase

38. An example of simple and permanent tissue is:
(a) Xylem (b) Phloem
(c) Selerenchyma (d) All the above

39. The shape of squamous epithelial tissue is:
(a) Cubical (b) Flattened
(c) Pillar like (d) None of the above

40. Fresh water sponge belongs to phylum:
(a) Arthropoda (b) Annelida

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