

## Limiting Reactants Worksheet KEY

Answer the following questions, using stoichiometry to convert between moles of one substance and moles of another substance within a reaction and using molar masses to convert between grams of a substance and moles of that substance.

Given the following reaction:  $\text{C}_5\text{H}_{12} + 8 \text{O}_2 \rightarrow 5 \text{CO}_2 + 6 \text{H}_2\text{O}$

1. If 5.25 grams of pentane ( $\text{C}_5\text{H}_{12}$ ) are completely reacted with oxygen ( $\text{O}_2$ ), how many grams of carbon dioxide ( $\text{CO}_2$ ) will be produced?

$$5.25 \text{ g C}_5\text{H}_{12} \times \frac{1 \text{ mol C}_5\text{H}_{12}}{72.151 \text{ g C}_5\text{H}_{12}} = 0.07276 \text{ mol C}_5\text{H}_{12}$$

$$0.07276 \text{ mol C}_5\text{H}_{12} \times \frac{5 \text{ mol CO}_2}{1 \text{ mol C}_5\text{H}_{12}} = 0.3638 \text{ mol CO}_2$$

$$0.3638 \text{ mol CO}_2 \times \frac{44.009 \text{ g CO}_2}{1 \text{ mol CO}_2} = 16.0 \text{ g CO}_2$$

2. If 9.75 grams of oxygen ( $\text{O}_2$ ) are completely reacted with pentane ( $\text{C}_5\text{H}_{12}$ ), how many grams of carbon dioxide ( $\text{CO}_2$ ) will be produced?

$$9.75 \text{ g O}_2 \times \frac{1 \text{ mol O}_2}{31.998 \text{ g O}_2} = 0.3047 \text{ mol O}_2$$

$$0.3047 \text{ mol O}_2 \times \frac{5 \text{ mol CO}_2}{8 \text{ mol O}_2} = 0.1904 \text{ mol CO}_2$$

$$0.1904 \text{ mol CO}_2 \times \frac{44.009 \text{ g CO}_2}{1 \text{ mol CO}_2} = 8.38 \text{ g CO}_2$$

3. Based on your answers to question #1 and 2, if 5.25 grams of pentane are reacted with 9.75 grams of oxygen, what is the limiting reactant and how many grams of  $\text{CO}_2$  actually form?

**Limiting reactant:  $\text{O}_2$**

**8.38 g  $\text{CO}_2$  actually form**

# Limiting Reactants Key

**Daniel W Foster Professor of Medical  
Ethics John Sadler**



## Limiting Reactants Key:

Mapping College Chemistry Stephen DeMeo, 2019-03-01 This text is a chemistry problem solving resource appropriate for teachers and their students who are enrolled in high school Advanced Placement Chemistry or in a first year college General Chemistry course The book incorporates a chemistry problem solving plan one that uses an innovative graphic organizer strategy The strategy successfully evaluated with students combines problem solving processes with chemical concepts that will allow students to solve the most common and difficult problems encountered in the first year of chemistry Topical problem solving will focus on limiting reactant stoichiometry identifying types of chemical reactions equilibrium acid base equilibria and electrochemistry Why would this resource be of interest to chemistry students To be successful to get into a well known college medical school physical therapy or graduate program often requires that students get an A in your pre requisite Introductory General Chemistry course To make matters worse many college professors feel that only a few students should get A grades and therefore they give difficult exams that many students fail this is the weeding out process that every pre health student is apprehensive about To succeed in this competitive environment entails not just studying harder or longer it means re organizing textbook content so that it is meaningful to the student This is the first text of its kind to employ a reliable research based strategy that incorporates a decision based visual tool to solve chemistry textbook problems ones that can make or break a career

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consumption accustomed to the use of cheap non renewable energy and to the acceptance of existing imbalances between the resources and demands of countries Despite all these stressing factors our world is still hesitating about the urgency of undertaking new and decisive research that could stabilize our future Could this trend change in the near future In our view two different scenarios are possible A renewed energy tension could take place with an unpredictable timing mostly related to political and economic factors This could bring again scientists and technologists to a new state of shock and awaken our talents A second interesting and beneficial scenario could result from the positive influence of a new generation of researchers that with or without immediate crisis acting both in industry and academia will face the challenge of developing technologies and processes to pave the way to a less vulnerable society Because Chemical Reactor Design and Technology activities are at the heart of these required new technologies the timeliness of the NATO Advanced Study Institute at the University of Western Ontario London was very appropriate

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