

TITRATIONS



Calculate the concentration of the sulfuric acid in mol/dm³

Use only the student's concordant results

mean titre = $\frac{22.10 + 22.15 + 22.15}{3} = 22.13 \text{ cm}^3$ ✓

$n(\text{NaOH}) = 0.105 \times (22.13/1000) = 0.002324 \text{ mol}$ ✓

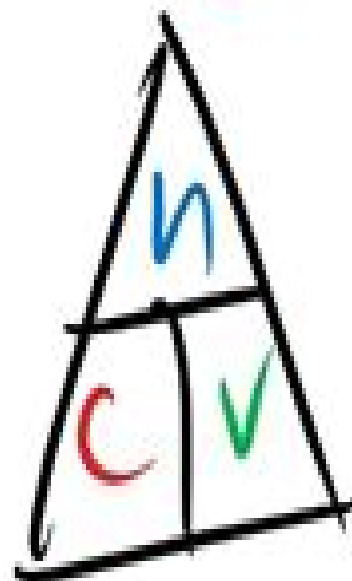
$n(\text{H}_2\text{SO}_4) = 0.002324 / 2 = 0.001162 \text{ mol}$ ✓

$c(\text{H}_2\text{SO}_4) = \frac{0.001162}{(25/1000)} = 0.0465 \text{ mol/dm}^3$ ✓

$$n = m/M_r$$

$$n = C \times V$$

$$C = n/V$$



GCSE CHEMISTRY

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Adam Robbins

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

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