


## QUESTION 2

	Test	Observation	Inference
(a)(i)	C + Conc. $H_2SO_4$ + strong heat	A copious brown irritating fume which turned moist blue litmus paper to red	Gas is $NO_2$ $NO_3^-$ present
(b)(i)	C + water + NaOH in drops, in excess	white gelatinous precipitate precipitate dissolved	$Al^{3+}$ or $Zn^{2+}$
(ii)	C + water + $NH_3$ in drops, in excess	white gelatinous precipitate precipitate dissolved	$Al^{3+}$ or $Zn^{2+}$
(iii)	C + water + freshly prepared $FeSO_4$ + Conc. $H_2SO_4$	A brown ring is formed at the junction of the acid and aqueous solution.	$NO_3^-$
(c)(i)	D + water	D dissolved	soluble
(ii)	Solution D + dil HCl	Colourless gas, smells like rotten egg, turns moist blue litmus paper to red and lead (II) ethanoate paper to black	$H_2S$ present
(iii)	Solution D + NaOH in drops, in excess	white precipitate precipitate insoluble	$Ca^{2+}$ or $Pb^{2+}$
(iv)	solution D + $(NH_4)_2CO_3$	white precipitate	$Ca^{2+}$

Test tube slightly warm

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