



**Western
Science**

Chemistry 1302B Final Exam *Answer Key*

Wednesday, April 19, 2023
2:00 – 5:00 pm

1. The density of the $\text{N}_2(\text{g})$ in a rigid tank at 250 kPa was 1.54 g L^{-1} . What was the temperature of the gas in this container?

547 K

2. Which of the following statements is true regarding the two samples below?

- Sample I: 3.0 moles of Ar at 20 °C in a container of volume V
- Sample II: 3.0 moles of Ar at 100 °C in a container of volume V

The pressure of Sample II is greater than that of Sample I.

3. Before contact with Europeans, Indigenous Peoples in this region produced maple syrup by evaporating the water from maple sap held in wooden containers. To heat the sap, they would drop in some stones that had been heated in a fire. If a 500 g stone that had been heated to 200 °C was added to a container holding 2.00 kg of sap at 5 °C, what temperature would the liquid have achieved when stone and sap reached thermal equilibrium? (Helpful data: The specific heat capacity of the stone was $0.790 \text{ J g}^{-1} \text{ K}^{-1}$ and that of the sap at this stage was the same as that of liquid water. Ignore heat losses to the container and surroundings.)

53.8 °C

4. Given: $\Delta H_f^\circ(\text{CO}_2\text{g}) = -130 \text{ kJ mol}^{-1}$, the ΔH° of the reaction $\text{CO}(\text{g}) + \text{H}_2\text{O}(\text{g}) \rightarrow \text{CO}_2(\text{g})$ is:

- I. ΔH_f° of $\text{CO}_2(\text{g})$
- II. The heat of combustion of $\text{CO}(\text{g})$
- III. -504 kJ

II only

5. Given: $\Delta H_{\text{sub}}^\circ = 402.6 \text{ kJ mol}^{-1}$ $\Delta H_{\text{fus}}^\circ = 535.6 \text{ kJ mol}^{-1}$

Estimate ΔH for the reaction: $\text{BF}_3(\text{g}) + \text{BCl}_3(\text{g}) \rightarrow \text{BF}_3\text{Cl}_3(\text{g}) + \text{BF}_2\text{Cl}_2(\text{g})$

0.0 kJ

6. How many of the processes below have a positive ΔS° ?

- $\text{N}_2(\text{g}) + 3 \text{H}_2(\text{g}) \rightarrow 2 \text{NH}_3(\text{g})$
- $2 \text{NaCl}(\text{s}) + \text{Cl}_2(\text{g}) \rightarrow 2 \text{NaCl}(\text{s})$
- Heating 10 g of water from 0 °C to 10 °C
- Condensing 10 g of water vapour into liquid at 100 °C

Gases Section 1 Answers

SJ Ball



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régime dissocié menu parfait et aliments pour perdre du poids - Sep 06 2023

web oct 25 2022 le régime dissocié consiste à ne pas mélanger plusieurs aliments au sein d'un même repas afin de favoriser la digestion tous les aliments peuvent être consommés à condition de ne pas être associés s'ils sont de nature différente en d'autres termes légumes et poisson ne peuvent pas cohabiter dans votre assiette au cours d'un même

alimentation dissociée wikipédia - Jan 30 2023

web l'alimentation dissociée est un terme désignant une approche nutritionnelle pseudo scientifique qui préconise des combinaisons spécifiques ou déconseille certaines combinaisons d'aliments certaines combinaisons sont présentées comme essentielles à une bonne santé à une meilleure digestion et à une perte de poids malgré l'