

7. (a) Yellow light of photon energy $3.5 \times 10^{-19} \text{ J}$ is incident on the surface of a particular photocell.



- (i) Calculate the wavelength of the light. [1]

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- (ii) Electrons emitted from the surface of the photocell have almost no kinetic energy. Explain why this does **not** contradict the law of conservation of energy. [2]

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- (b) Radiation of photon energy $5.2 \times 10^{-19} \text{ J}$ is now incident on the photocell. Calculate the maximum velocity of the emitted electrons. [2]

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(This question continues on the following page)



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Michael Rajnik



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