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Source: <http://www.fishbase.org>.

10. A sample of gas at 4.0°C and 1.00 atmosphere occupies 2.00 L. What volume will this gas occupy at 10°C and 1.00 atm?
11. A sample of hydrogen gas (H₂) has a volume of 0.50 L at a temperature of 0°C and a pressure of 1.00 atm. Calculate the volume of H₂ molecules present inside the sample.
12. An amount of substance (n) mol. of compressed gas at 1.00 atm pressure. Inside which this gas is trapped into a rigid plastic bag, the bag inflates to a volume of 2.00 L. What is the pressure of the gas inside the plastic bag? Assume the temperature is constant.
13. A 0.50 L gas sample at 10°C and a pressure of 101.7 kPa expands to a volume of 0.75 L. What pressure of the gas is 101.7 kPa? What is the final temperature of the gas in degrees Celsius?
14. What is the volume of 0.75 moles of H₂ at 10°C?
15. A substance filled in a volume of 1.00 m³ at a temperature of 10.0°C. This substance expanded at constant pressure to a temperature of 1.00°C (10°C). What is the final volume of the substance?
16. A 0.500 mol of nitrogen gas occupies a volume of 0.50 L at 0°C. What volume will 0.500 mol of nitrogen gas occupy at the same temperature and pressure?
17. A 0.50 L sample is filled with 0.01 g of gas. What is the temperature of the pressure in 0°C unit?
18. A particular substance is designed for its application to be filled to a volume of no more than 1.00 L. If the substance is filled into 0.50 L of container now filled (1.00 atm), the container will burst at a volume of 1.00 L. What is the temperature of the gas in 0°C unit? (Assume the substance is ideal)
19. A compressed gas cylinder at 101.7 kPa and 10°C. It has a mass of 10.0 kg. What is the temperature in 0°C? What is the final pressure in the cylinder?
20. A cylinder at 10°C is a volume of 0.50 L. It has 0.01 g of H₂ gas and 0.01 g of O₂ gas. Calculate the total pressure (in kPa) for the cylinder.
21. What volume is occupied by 0.75 mol of a diatomic gas at 10°C?
22. Compare the diffusion rates of H₂ and N₂ gases.
23. Three moles of a substance at 10.0 L, 10.0 atm and 10.0°C.

Gases Study Guide Answers

Lei Huang

A red circular graphic with a gradient, appearing as a stylized arrow or a partial circle, located to the right of the author's name.

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