

## Chapter 14—WS GAS LAWS

1. A quantity of gas under a pressure of 800 mm Hg has a volume of 380 liters. What is its volume at standard pressure when temperature is held constant?
2. A quantity of gas has a volume of 120 liters when confined under a pressure of 700.0 mm Hg at a temperature of 20.0°C. At what pressure will its volume be 30.0 liters at 20.0°C?
3. At constant pressure, the volume of a gas is increased from 150 liters to 300.0 liters by heating it. If the original temperature of the gas was 20.0°C, what will its final temperature be?
4. A quantity of gas exerts a pressure of 740 mm Hg at a temperature of 22.0°C. If the volume remains unchanged, what pressure will it exert at -8.0°C?
5. A quantity of oxygen gas has a volume of 850 ml when measured at 27.0°C and 730 mm Hg of pressure. Determine its volume at STP.
6. When measured at STP, a quantity of gas has a volume of 500.0 liters. What volume will it occupy at 0.0°C and a pressure of 700.0 mm Hg?
7. Know the relationships between your four variables- pressure, volume, temperature, and moles. Which are inverse and which are direct?? Be able to explain them in terms of what is happening to the energy, collisions, and space between the particles.
8. A gas whose behavior closely resembles that of an ideal gas has a volume of 2.00 liters at a temperature of 27.0°C and a pressure of 900.0 mm Hg. (a) How many moles of molecules are in the sample? (b) How many molecules are in the sample?
9. Know Graham's law of effusion. Be able to calculate the ratio of rate of effusion. Example: what is the rate of effusion of oxygen gas compared to nitrogen if both gases are at the same temperature?

10. How many moles?

$$1. P_1 V_1 = P_2 V_2$$

$$800 \text{ mmHg} (380 \text{ L}) = 760 \text{ mmHg} (V_2)$$

$$V_2 = 400 \text{ L}$$

$P \downarrow V \uparrow$  checks out!

$$2. P_1 V_1 = P_2 V_2$$

$$700.0 \text{ mmHg} (120 \text{ L}) = P_2 (30 \text{ L})$$

$$P_2 = 2800 \text{ mmHg}$$

$V \downarrow P \uparrow$  checks out!

$$3. \frac{V_1}{T_1} = \frac{V_2}{T_2} \quad \frac{150 \text{ L}}{293 \text{ K}} = \frac{300 \text{ L}}{T_2}$$

$$T_2 (150) = 300 (293)$$

$$T_2 = 586 \text{ K}$$

$V \uparrow T \uparrow$   
checks out!

# Gas Laws Review Sheet Answers

**Yan Bai**

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the name 'Yan Bai'.

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Kenneth W. Raymond, 2005-10-07 Finally readers have a shorter less intimidating introduction to general organic and biological chemistry Not only is Raymond's text concise it also takes an integrated approach to presenting important topics in a way that makes the material easier to understand In this approach similarities can be exploited and concepts reinforced The result is that readers see the strong connections that exist between these three branches of chemistry Elgar Concise Encyclopedia of Oil and Gas Law Tina Soliman Hunter, Angela Dobb, Ernst Nordtveit, 2024-11-08 The Elgar Concise Encyclopedia of Oil and Gas Law provides a comprehensive overview of the engineering and geological aspects of oil and gas activities placed within their legal context as well as legal aspects of these activities It focuses on exploration for and production of oil and gas incorporating experience based knowledge and the application of the law to technical issues

*General, Organic, and Biological Chemistry Study Guide and Selected Solutions* Karen C. Timberlake, 2001-11 Keyed to the learning goals in the text this guide is designed to promote active learning through a variety of exercises with answers and mastery exams The guide also contains complete solutions to odd numbered problems **Harvard Law Review:**

**Volume 129, Number 4 - February 2016** Harvard Law Review, 2016-02-09 The February 2016 issue Number 4 features these contents Article Constitutional Bad Faith by David E Pozen Book Review No Immunity Race Class and Civil Liberties in Times of Health Crisis by Michele Goodwin FTC regulation of data security voting rights disparate impact and the Texas voter ID law and fair labor primary beneficiary and unpaid interns The issue includes analysis of Recent Regulations on Dodd Frank and mandatory pay disclosure and on Clean Air Act regulation of carbon emissions from existing power plants Also included are a Recent Event comment on the killing of a non university affiliate by campus police and a Recent Book comment on Richard McAdams 2015 book The Expressive Powers of Law Finally the issue includes several brief comments on Recent Publications The Harvard Law Review is offered in a quality digital edition featuring active Contents linked footnotes active URLs legible tables and proper ebook and Bluebook formatting The Review is a student run organization whose primary purpose is to publish a journal of legal scholarship It comes out monthly from November through June and has roughly 2500 pages per volume Student editors make all editorial and organizational decisions This is the fourth issue of academic year 2015 2016 *AP Chemistry For Dummies* Peter J. Mikulecky, Michelle Rose Gilman, Kate Brutlag, 2008-11-13

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