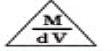


## Density Calculations Worksheet I



Find the unknown quantity:

a)	d = 3 g/mL
	V = 100 mL
	M = ?
	$m = d \times v$
	= (3glmL)(100mL)
	= 3 × 10 2 3

0) d = ?  
V = 950 mL  
M = 95 g  
d = 
$$\frac{m}{V}$$
  
=  $\frac{959}{750mL}$  = 0.10g/mL

= ?  
= 950 mL  
I = 95 g  
= 
$$\frac{M}{V}$$
  
=  $\frac{959}{750mL}$  = 0.109 mL  
C) d = 0.5 g/cm<sup>3</sup>  
V = ?  
M = 20 g  
V =  $\frac{M}{V}$   
=  $\frac{209}{0.59 \text{ cm}^3}$  = 4× 10 cm<sup>3</sup>

Find the unknown quantity (CONVERT FIRST to g or mL)

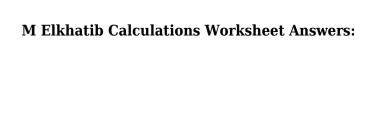
#### WORD PROBLEMS

A block of aluminum occupies a volume of 15.0 mL and weighs 40.5 g. What is its density?

Mercury metal is poured into a graduated cylinder that holds exactly 22.5 mL. The mercury used to fill the cylinder weighs 306.0 g. From this information, calculate the density of mercury.

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