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This question is not based on a passage.

**Question 4**

Which of these statements concerning peptide bonds is FALSE?

- ☐ (A) Their formation involves a reaction between an amino group and a carboxyl group.
- ☐ (B) They are the primary bonds that hold amino acids together.
- ☐ (C) They have partial double bond character.
- ☒ (D) Their formation involves hydrolytic reactions.

 incorrect

**B**

Peptide bonds are the primary covalent bond between the amino acids that make up proteins, making choice (B) incorrect. They involve a condensation reaction between the amino group of one amino acid and the carboxyl group of an adjacent amino acid, eliminating choice (A). The peptide bond has a partial double bond character because the double bond can resonate between C=O and C=N. Thus, the peptide bond has a partial double bond character and exhibits limited rotation, eliminating choice (C). By process of elimination, choice (D) is false. Formation of the peptide bond is a condensation reaction—specifically a dehydration reaction involving the loss of water—not a hydrolytic reaction involving the addition of water.

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# Kaplan Explanations Aamc Test 10

**Puja Mehta**



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### **Table of Contents Kaplan Explanations Aamc Test 10**

1. Understanding the eBook Kaplan Explanations Aamc Test 10
  - The Rise of Digital Reading Kaplan Explanations Aamc Test 10
  - Advantages of eBooks Over Traditional Books
2. Identifying Kaplan Explanations Aamc Test 10
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Kaplan Explanations Aamc Test 10
  - User-Friendly Interface
4. Exploring eBook Recommendations from Kaplan Explanations Aamc Test 10
  - Personalized Recommendations
  - Kaplan Explanations Aamc Test 10 User Reviews and Ratings
  - Kaplan Explanations Aamc Test 10 and Bestseller Lists

5. Accessing Kaplan Explanations Aamc Test 10 Free and Paid eBooks
  - Kaplan Explanations Aamc Test 10 Public Domain eBooks
  - Kaplan Explanations Aamc Test 10 eBook Subscription Services
  - Kaplan Explanations Aamc Test 10 Budget-Friendly Options
6. Navigating Kaplan Explanations Aamc Test 10 eBook Formats
  - ePub, PDF, MOBI, and More
  - Kaplan Explanations Aamc Test 10 Compatibility with Devices
  - Kaplan Explanations Aamc Test 10 Enhanced eBook Features
7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Kaplan Explanations Aamc Test 10
  - Highlighting and Note-Taking Kaplan Explanations Aamc Test 10
  - Interactive Elements Kaplan Explanations Aamc Test 10
8. Staying Engaged with Kaplan Explanations Aamc Test 10
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Kaplan Explanations Aamc Test 10
9. Balancing eBooks and Physical Books Kaplan Explanations Aamc Test 10
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Kaplan Explanations Aamc Test 10
10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine Kaplan Explanations Aamc Test 10
  - Setting Reading Goals Kaplan Explanations Aamc Test 10
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Kaplan Explanations Aamc Test 10
  - Fact-Checking eBook Content of Kaplan Explanations Aamc Test 10
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

#### 14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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