II. Physical Properties of Solutions

- a. Solute—substance that gets dissolved or dispersed in a solution; present in a smaller amount; dispersed throughout the solvent, occupy positions that are normally taken by solvent molecules
- b. Solvent—the substance that does the dissolving; present in a larger amount; phase is retained in solution (aka dissolving medium)
- c. Solubility— how much solute will dissolve in a solvent at a specific temp; soln are characterized by their capacity to do so

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Robert R. Redfield

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