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FOR THE AUSTRALIAN CURRICULUM

STUDENT WORKBOOK

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 $\text{Br}_2 + \text{Fe}_2\text{O}_3 \rightarrow \text{FeBr}_3 + \text{O}_2$
 $\text{C}_6\text{H}_{10} + \text{O}_2 \rightarrow \text{CO}_2 + \text{H}_2\text{O}$
 $\text{H}_2\text{O} + \text{KBr} + \text{CO}_2 \rightarrow \text{GaBr}_3 + \text{Na}_2\text{SO}_3 + \text{Ga}_2\text{SO}_3 + \text{NaBr}$
 $\text{SNO}_2 + \text{NF}_3 \rightarrow \text{SNF}_2 + \text{N}_2\text{O}_3$
 using the following equation
 $\text{NaOH} + \text{H}_2\text{SO}_4 \rightarrow \text{H}_2\text{O} + \text{Na}_2\text{SO}_4$

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 $\text{GaBr}_3 + \text{Na}_2\text{SO}_3 \rightarrow \text{Ga}_2\text{SO}_3 + \text{NaBr}$
 $\text{SNO}_2 + \text{NF}_3 \rightarrow \text{SNF}_2$

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 $\text{SiO}_2 + 3\text{C} \rightarrow \text{SiC} + 2\text{CO}$

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 $\text{HBr} + \text{KHCO}_3 \rightarrow \text{H}_2\text{O} + \text{KBr} + \text{CO}_2$
 $\text{GaBr}_3 + \text{Na}_2\text{SO}_3 \rightarrow \text{Ga}_2\text{SO}_3 + \text{NaBr}$
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