

PART - I

15 X 1 = 15

- I Answer all the questions.**
- Which of the following is / are true with respect to Carbon - 12  
a) relative atomic mass is 12u      b) oxidation number of Carbon is +4 in all its compounds      c) mole of Carbon - 12 contain  $6.022 \times 10^{22}$  carbon atoms      d) all of these
  - How many electrons in an atom with atomic number 105 can have  $(n + l) = 8$   
a) 30      b) 17      c) 15      d) unpredictable
  - Among the following which is isoelectronic with  $O^{2-}$  ion,  
a)  $Cl^-$       b)  $F^-$       c)  $Ca^{2+}$       d)  $N^{2-}$
  - The cause of permanent hardness of water is due to  
a)  $Ca(HCO_3)_2$       b)  $Mg(HCO_3)_2$       c)  $CaCl_2$       d)  $MgCO_3$
  - Sodium is stored in  
a) Alcohol      b) Water      c) Kerosene      d) None of these.
  - It temperature and volume of an ideal gas is increased to twice its values, the initial pressure P becomes  
a) 4p      b) 2p      c) p      d) 3p
  - Heat of combustion is always  
a) Positive      b) negative      c) zero      d) either positive or negative
  - In the reaction,  $Fe(OH)_3(s) \rightleftharpoons Fe^{3+}(aq) + 3OH^-(aq)$  if the concentration of  $OH^-$  ions is decreased by  $1/4$  times, then the equilibrium concentration of  $Fe^{3+}$  will  
a) not changed      b) also decreased by  $1/4$  times  
c) increase by 4 times      d) increase by 64 times
  - According to Raoult's law, the relative lowering of vapour pressure for a solution is equal to  
a) mole fraction of solvent      b) mole fraction of solute  
c) number of moles of solute      d) number of moles of solvent
  - Number of sigma ( $\sigma$ ) and pi ( $\pi$ ) bonds in acetylene is .....  
a)  $2\sigma$  &  $2\pi$       b)  $3\sigma$  &  $3\pi$       c)  $2\sigma$  &  $3\pi$       d)  $3\sigma$  &  $2\pi$
  - Sodium nitropruside reacts with sulphide ion to give a purple colour due to the formation of  
a)  $[Fe(CN)_5 NO]^{3-}$       b)  $[Fe(NO)_5 CN]^+$       c)  $[Fe(CN)_5 NOS]^{4-}$       d)  $[Fe(CN)_5 NOS]^{3-}$
  - What is the hybridisation state of benzyl carbonium ion?  
a)  $sp^2$       b)  $sp^3$       c)  $sp^2$       d)  $sp^2d$
  - ..... is used for welding and cutting metals.  
a) Ethylene      b) Propylene      c) Poly ethylene      d) Oxy - acetylene
  - What should be the correct IUPAC name of diethyl chloromethane?  
a) 3 - chloro pentane      b) 1 - chloropentane  
c) 1 - chloro - 1, 1, diethyl methane      d) 1 - chloro - 1 - ethylepropane
  - The pH of normal rain water is .....  
a) 6.5      b) 7.5      c) 5.6      d) 4.6

PART - II

6 x 2 = 12

- II Answer any six. (q.No. 24 is compulsory).**
- Define modern periodic law.
  - What is water - gas shift reaction?

# Grade 1chemistry Half Yearly Exam

**David Kirk**



## **Grade 1 chemistry Half Yearly Exam:**

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**Parliamentary Papers** Great Britain. Parliament. House of Commons,1868 *UCSF School of Pharmacy Bulletin* University of California, San Francisco. School of Pharmacy,1979 **New Developments in Pathways Towards Diversity and Inclusion in STEM: A United States Perspective** Alexander Gates,Juan Gilbert,Chris Botanga,Kim Nguyen,Bonita London,2022-10-17 The Louis Stokes Alliances for Minority Participation LSAMP program of the US National Science Foundation has been a primary force for raising the success and graduation of minority students in STEM for 30 years Increasing the number of underrepresented students earning baccalaureate degrees and entering graduate school in STEM is the goal of LSAMP This goal has been nearly achieved through the formation of alliances of degree granting institutions of higher learning varying from community colleges to major research institutions Currently there are 59 alliances including more than 400 institutions LSAMP is responsible for more than 650 000 bachelor s degrees earned by minority students in STEM The papers for this Research Topic should focus on the use of LSAMP activities programs and collaborations to develop pathways to success and graduation of STEM majors from minority groups that underrepresented in STEM These pathways can include any segment from pre college through graduate school Areas of special interest include mentoring research experiences transitions between levels and novel approaches for retention The studies should be research based and rigorous They can be pure research studies curriculum and design or literature reviews but they must be at a cutting edge level and be subject to detailed review and assessment *Journal of Gas Lighting and Water Supply* ,1892 **Schools Inquiry Commission: ... Special reports of assistant commissioners, and digests of information received** Great Britain. Schools Inquiry Commission,1868 Catalogue Number Johns Hopkins University. School of Engineering,1922 **School Science and Mathematics** ,1925 **Report of the commissioners** Schools inquiry commission,1869 *The Journal of Gas Lighting, Water Supply & Sanitary Improvement* ,1892 **Catalogue of Columbia University** Columbia University,1916 **Schools Inquiry Commission** ,1868 **Journal of Chemical Education** ,1928

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