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⋘Elements & Macromolecules in Organisms

Most common elements in living things are carbon, hydrogen, nitrogen, and oxygen. These four elements constitute about 95% of your body weight. All compounds can be classified in two broad categories — organic and inorganic compounds. Organic compounds are made primarily of carbon. Carbon has four outer electrons and can form four bonds. Carbon can form single bonds with another atom and also bond to other carbon molecules forming double, triple, or quadruple bonds. Organic compounds also contain hydrogen. Since hydrogen has only one electron, it can form only single bonds.

Each small organic molecule can be a unit of a large organic molecule called a macromolecule. There are four classes of macromolecules (polysaccharides or carbohydrates, triglycerides or lipids, polypeptides or proteins, and nucleic acids such as DNA & RNA). Carbohydrates and lipids are made of only carbon, hydrogen, and oxygen (CHO). Proteins are made of carbon, hydrogen, oxygen, and nitrogen (CHON). Nucleic acids such as DNA and RNA contain carbon, hydrogen, oxygen, nitrogen, and phosphorus (CHON P).

Use the drawing of the **amino acid on this worksheet** (look ahead to another page for this sketch and remember that the NUMBER OF LINES from a single atom is their NUMBER OF BONDS) to determine the number of bonds formed by:

_____ Oxygen _____ Hydrogen _____ Nitrogen

The body also needs trace amounts of other elements such as calcium, potassium, and sulfur for proper functioning of muscles, nerves, etc.

Questions:

- 1. Name the 4 main elements that make up 95% of an organism.
- 2. Name the 4 types of bonds carbon can form.
- 3. What are macromolecules?
- 4. Name the 4 classes of macromolecules.
- 5. Give 2 examples of nucleic acids. _______ 6. What elements make up carbohydrates & lipids (symbols)? ______
- 7. Name 3 elements your body needs trace amounts of for proper functioning.

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Table of Contents Elements And Macromolecules In Organisms Packet

- 1. Understanding the eBook Elements And Macromolecules In Organisms Packet
 - \circ The Rise of Digital Reading Elements And Macromolecules In Organisms Packet
 - Advantages of eBooks Over Traditional Books
- 2. Identifying Elements And Macromolecules In Organisms Packet
 - $\circ \ Exploring \ Different \ Genres$
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Elements And Macromolecules In Organisms Packet
 - User-Friendly Interface
- 4. Exploring eBook Recommendations from Elements And Macromolecules In Organisms Packet

- Personalized Recommendations
- Elements And Macromolecules In Organisms Packet User Reviews and Ratings
- Elements And Macromolecules In Organisms Packet and Bestseller Lists
- 5. Accessing Elements And Macromolecules In Organisms Packet Free and Paid eBooks
 - Elements And Macromolecules In Organisms Packet Public Domain eBooks
 - Elements And Macromolecules In Organisms Packet eBook Subscription Services
 - Elements And Macromolecules In Organisms Packet Budget-Friendly Options
- 6. Navigating Elements And Macromolecules In Organisms Packet eBook Formats
 - o ePub, PDF, MOBI, and More
 - Elements And Macromolecules In Organisms Packet Compatibility with Devices
 - Elements And Macromolecules In Organisms Packet Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Elements And Macromolecules In Organisms Packet
 - Highlighting and Note-Taking Elements And Macromolecules In Organisms Packet
 - Interactive Elements Elements And Macromolecules In Organisms Packet
- 8. Staying Engaged with Elements And Macromolecules In Organisms Packet
 - o Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - \circ Following Authors and Publishers Elements And Macromolecules In Organisms Packet
- 9. Balancing eBooks and Physical Books Elements And Macromolecules In Organisms Packet
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Elements And Macromolecules In Organisms Packet
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Elements And Macromolecules In Organisms Packet
 - Setting Reading Goals Elements And Macromolecules In Organisms Packet
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Elements And Macromolecules In Organisms Packet

- Fact-Checking eBook Content of Elements And Macromolecules In Organisms Packet
- Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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