

Electronegativity and Bond Polarity

WHY?

Electronegativity is a measure of the ability of an atom in a molecule to attract electrons. The difference in the electronegativities of two atoms profoundly affects the properties of the chemical bond between the two atoms and, consequently, has dramatic effects on the physical and chemical properties of materials. You therefore need to be able to identify polar bonds and polar molecules from atomic electronegativities and molecular geometries (i.e., shapes).

LEARNING OBJECTIVES

- Master the use of electronegativity to predict bond characteristics
- Identify how bond characteristics affect material properties

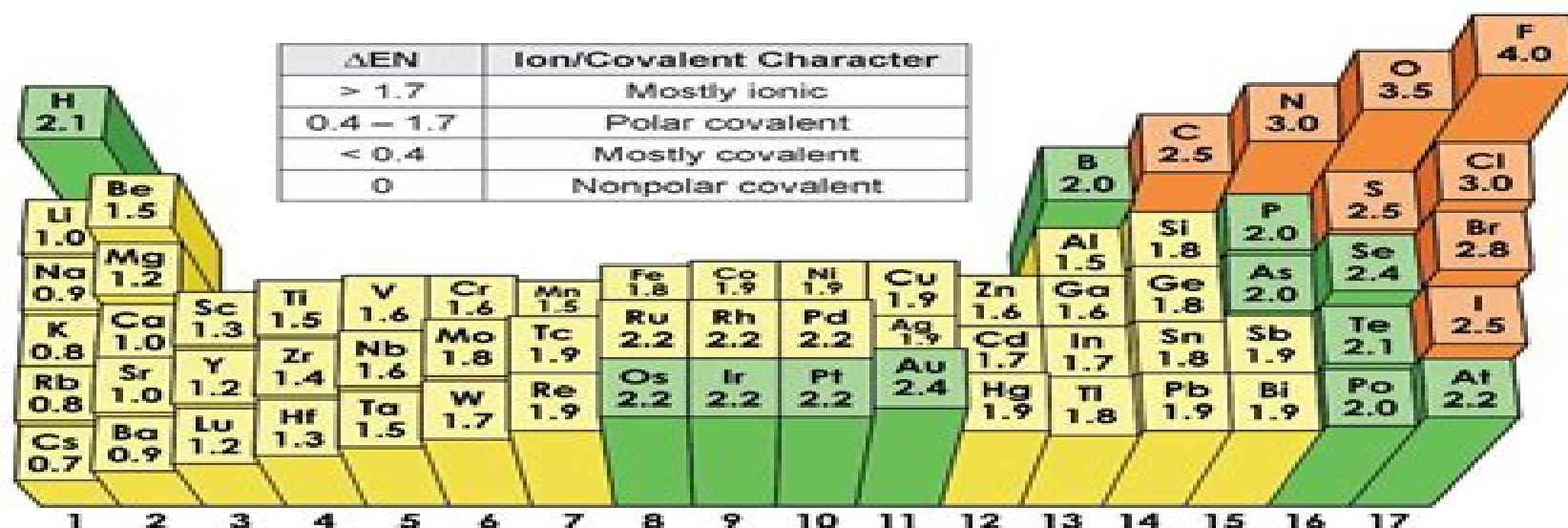
SUCCESS CRITERIA

- Correctly identify bonds in order of increasing polarity
- Correctly identify polar molecules

PREREQUISITES

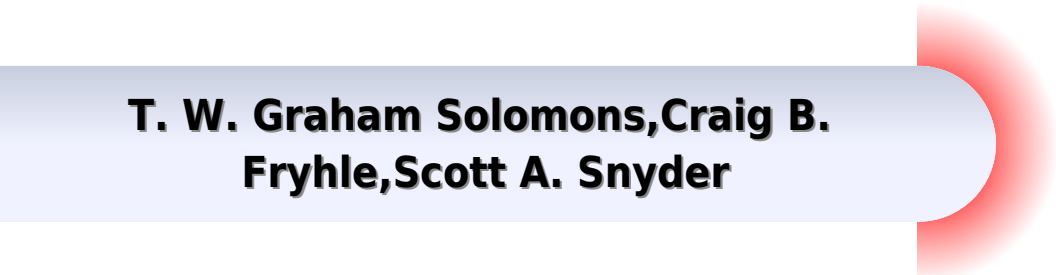
- Activity 07-5: *Periodic Trends in Atomic Properties*
- Activity 09-1: *Valence Shell Electron Pair Repulsion Model*

MODEL: ELECTRONEGATIVITY (EN) AND BOND CHARACTERISTICS



Electronegativity And Polarity Study Guide Answers

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Fryhle, Scott A. Snyder**



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