LABORATORY MODULE 3 SIMPLE STAIN



Visualizing microorganisms in the living state is quite difficult, not only because they are minute, but also because they are manaparent and practically colorless when suspended in an aqueous medium. To study their properties and to divide microorganisms into specific groups for diagnostic purposes, biological stains and staining procedures in conjunction with light microscopy have become major tools in microscopy. Chemically, a stein (dye) is an organic compound containing a bencome ring plus a chromophore and an auxochrome group.



Figure 3.1. Chemical composition of a stain.

The ability of a state to bind to exectomolocular cellular components such as promise or nucleic acids depends on the electrical charge found on the chromogen portion, as well as on the cellular component to be stained.

Acidic states are actionic, which means that, on ionization of the state, the absonages portion exhibits a regative charge and therefore has a strong affinity for the positive constituents of the cell. Proteins, positively charged cellular components, will readily bind to and accept the color of the negatively charged, anionic chromogen of an acidic state. Structurally, pionic acid is an example of an acidic state that produces an anionic chromogen.



Figure 3: 2: Prono solot an acretic stain.

Basic stains are cationic, because on tonization the chromogen portion exhibits a positive charge and therefore has a strong affinity for the negative constituents of the cell. Nucleic acids, negatively charged cellular components, will readily bind to and accept the color of the positively charged, carionic chromogen of a basic.

Lab Module Gram Staining Introduction Clark College

SJ Ball

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The Gram Stain Gilda L. Jones,1986

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