

ঘের ও কীটপতঙ্গের অস্বাস্থ্যের নির্বাচনী পরীক্ষার প্রয়োজক

চলকের ঘনমাত্রা,

$$S = \frac{W}{V} = \frac{20}{36.5} \times 1$$

$$M = 0.5479 \text{ M}$$

অতএব, HCl চলকের ঘনমাত্রা

$$= 0.5479 \text{ M}$$

এখানে,

চলকের ভর, $m = 20 \text{ g}$

HCl এর আপেক্ষিক ভর,

$$M = 36.5 \text{ g/mol}$$

চলকের আয়তন, $V = 1 \text{ L}$

চলকের ঘনমাত্রা $S = ?$

উদাহরণে উল্লিখিত N -পদ্ধতিতে 10 mg এর মধ্যে কয় পরিমাণ চূর্ণ ও V -পদ্ধতিতে 0.2 g 2% ($\frac{W}{V}$) HCl চূর্ণের সাথে মিশ্রিত করা হয়েছে।

যদিও N -পদ্ধতিটি হলো সেমি-মাইক্রো আনালিটিক্যাল পদ্ধতি ও V -পদ্ধতিটি হলো মাইক্রো আনালিটিক্যাল পদ্ধতি। উভয় পদ্ধতির মধ্যে সেমি-মাইক্রো পদ্ধতি পরিবেশে অস্বাস্থ্য। নিম্নে তা বিশ্লেষণ করা হলো-

সেমি-মাইক্রো পদ্ধতি রাসায়নিক গুণগত বিশ্লেষণের একটি পদ্ধতি। এটি বিশ্লেষণের সময় যথেষ্ট রাসায়নিক উপাদান সন্নিবেশিত ও বিশুদ্ধতায় বিশুদ্ধতায় ব্যবহৃত হয়। এ পদ্ধতিতে খুব কম পরিমাণ (0.05 থেকে 0.2 g কঠিন বা $2-4 \text{ mL}$ তরল দ্রব্য) রাসায়নিক পদার্থ নিয়ে পরীক্ষা করা হয়। ফলে পরিবেশে মুদ্রণ উল্লেখযোগ্যভাবে কমে আসে।

এছাড়া সেমি-মাইক্রো পদ্ধতিতে H_2SO_4 এর পরিমাণে কয়েকটিমাইক্রো ব্যবহার করা হয় ফলে পরিবেশে মুদ্রণ ঘটে না।

$(\text{CN})_2 + \text{H}_2\text{O} \rightarrow \text{HCN} + \text{HCN}$

অতএব, কোনো কোনো ক্ষেত্রে কম কৃত্রিম রাসায়নিক দ্রব্য ব্যবহৃত হয়। ফলে পরিবেশে মুদ্রণের সম্ভাবনা কমে আসে। তাই সেমি-মাইক্রো পদ্ধতি পরিবেশে অস্বাস্থ্য।

চিত্র-১০



১. টাইটার ফ্লাস্ক কী?
২. সেমি-মাইক্রো পদ্ধতি পরিবেশে অস্বাস্থ্য কেন?
৩. চলকের জন্য প্রয়োজ্য হ্যাঙ্গার প্রতীকটি অঙ্কন করে এবং ব্যাখ্যা করে।
৪. উদাহরণের অস্বাস্থ্যকে চূর্ণ থেকে উপযুক্ত কাচ পাত্র ব্যবহার করে 1 M ঘনমাত্রার চূর্ণ তৈরি করার প্রক্রিয়া ব্যাখ্যা করে। ৪

৫. বিশ্লেষণের নিমিত্ত বীমের উপর টাইটার স্থাপন করলে বীমের প্রতি পাশের ভর সমান হলে চূর্ণের ঘনমাত্রা, তাকে টাইটার ফ্লাস্ক বলে।

৬. সেমি-মাইক্রো পদ্ধতি রাসায়নিক গুণগত বিশ্লেষণের একটি পদ্ধতি। এটি বিশ্লেষণের সময় যথেষ্ট রাসায়নিক উপাদান সন্নিবেশিত ও বিশুদ্ধতায় বিশুদ্ধতায় ব্যবহৃত হয়। এ পদ্ধতিতে খুব কম পরিমাণ (0.05 থেকে 0.2 g কঠিন বা $2-4 \text{ mL}$ তরল দ্রব্য) রাসায়নিক পদার্থ নিয়ে পরীক্ষা করা হয়। ফলে পরিবেশে মুদ্রণ উল্লেখযোগ্যভাবে কমে আসে।

৭. অতএব, কোনো কোনো ক্ষেত্রে কম কৃত্রিম রাসায়নিক দ্রব্য ব্যবহৃত হয়। ফলে পরিবেশে মুদ্রণের সম্ভাবনা কমে আসে। তাই সেমি-মাইক্রো পদ্ধতি পরিবেশে অস্বাস্থ্য।

প্রতীক	অর্থ (Corrosive)
আমদান	অস্বাস্থ্য রাসায়নিক উপাদান কলতে যে ক্ষয়জনক রাসায়নিক দ্রব্যকে চূর্ণের সাথে মিশ্রিত করে ও অস্বাস্থ্য পাত্রের মাধ্যমে কলিত পদ্ধতি হয়। এতে তরল পুড়ে যায় এবং অস্বাস্থ্য পাত্রের পুড়ে যায়।
উদাহরণ	পাত্র কলিত এমিউলসন (HCl , HNO_3 , H_2SO_4) পাত্র কলিত (NaOH , KOH , H_2O_2) আকস্মিক লিকার প্রভৃতি চূর্ণের উপস্থিতি।
সংরক্ষণ পদ্ধতি	<ul style="list-style-type: none"> পাত্র এমিউলসে পোতলে রাখতে হবে। পাত্র কলিতে, কলিত পাত্রের পুড়ে পুড়ে পুড়ে রাখতে হবে।

৮. উদাহরণের মূলটি হলো 10 M H_2SO_4 চূর্ণ এবং এর থেকে 1 M ঘনমাত্রার চূর্ণ প্রস্তুত করতে হবে 1 M ও 1 M কলিত দ্রব্যের কলিত হবে।

1 M কলিত দ্রব্য হলো 100 mL আকস্মিক রাসায়নিক চূর্ণ ও এর টি হলো পিপিটি। সেমি-মাইক্রো রাসায়নিক চূর্ণের আয়তন 100 mL । তাই আমাদের 100 mL 1 M ঘনমাত্রার H_2SO_4 চূর্ণ তৈরি করতে হবে।

চূর্ণের লব্ধিকরণ পুত্র হতে জানতে পারি,

$$V_1 S_1 = V_2 S_2$$

$$\text{বা, } V_1 = \frac{V_2 S_2}{S_1}$$

$$= \left(\frac{100 \times 1}{10} \right) \text{ mL}$$

$$= 10 \text{ mL}$$

অতএব, 10 mL H_2SO_4 চূর্ণের সঙ্গে পানি যোগ করতে হবে

$(100 - 10) \text{ mL} = 90 \text{ mL}$

অতএব, পিপিটির মাধ্যমে 10 mL চূর্ণের ঘনমাত্রার চূর্ণ নিয়ে তার সাথে আকস্মিক রাসায়নিক চূর্ণ 90 mL পানি যোগ করলে আমাদের কলিত ঘনমাত্রার চূর্ণ পাওয়া যাবে।

চিত্র-১১



১. অস্বাস্থ্য রাসায়নিক কলিতে কলি?
২. সাদা চূর্ণ $\text{K}_2\text{Cr}_2\text{O}_7$ এর ঘন কলিতে অস্বাস্থ্য হয় না-ব্যাখ্যা করে।
৩. A চূর্ণের হ্যাঙ্গারের উত্তর করার কৌশল বর্ণনা করে।
৪. B ও C চূর্ণের সাথে সম্পর্কিত রাসায়নিক চূর্ণের ব্যবহারকালে খাওয়া কৃত্রিম চূর্ণের উপর সম্পর্কিত বিশ্লেষণ করে। ৪

৫. ৫ নং প্রশ্নের উত্তর

৬. অস্বাস্থ্য রাসায়নিক বা কৃত্রিম দ্রব্য লব্ধিকরণ বা মুদ্রণ বা উপাদানসমূহকে জারণ প্রক্রিয়ায় জারণ থেকে রক্ষা করে রাখার সংরক্ষণ করে।

৭. সামান্য $\text{K}_2\text{Cr}_2\text{O}_7$ এর ঘন কলিতে অস্বাস্থ্য বা শূন্য হতে পারে না। উত্তরটি ব্যাখ্যা করার জন্য নিম্নের উত্তরসূরী মাইক্রোপরি বিশ্লেষণ করি-

$$A + B \rightarrow C + D$$

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N Colangelo



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