

Lab 11

Gravimetric Analysis

Pre-Lab Questions

1. Write a balanced equation for the precipitation of calcium carbonate from potassium carbonate and calcium chloride.
$$\text{K}_2\text{CO}_3 + \text{CaCl}_2 \rightarrow \text{CaCO}_3 + 2\text{KCl}$$
2. Using this balanced equation, determine the limiting reactant if 15 grams of calcium chloride was reacted with 15 grams of potassium carbonate.

Using your answer for Question 2, determine the mass of potassium carbonate needed to fully precipitate all the calcium from a 25 mL sample of 15% calcium chloride.

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