

# ANSWER KEY

## ATOMS AND ELECTRONS STUDY GUIDE

**Learning Goals:** (a) (a.1-P.4.1) Explain a new scientific theory of atomic-scale phenomena using theory by describing changes in the atomic model over time, including those changes that are inconsistent with experimental evidence. (a.1-P.4.2) Explain the scientific theory of atomic-scale phenomena by describing the scientific changes in terms of changing scientific understandings, including scientific theory that are not fully or some of their more startling changes, and evidence within the atom. (a.1-P.4.3) Explain properties of atoms and their patterns in the periodic table, including the periodicity of their properties, related to P.4.1, as changes in the scientific understandings of their fundamental properties (including periodicity, electronegativity, ionization energy, atomic mass, etc.).

### ANSWER KEY

**Directions:** Match the atomic theory selected on the correct understanding.

| Model/Model | Scientist(s)        | Atomic Theory/Atomic Model Theory                                                                                                  |
|-------------|---------------------|------------------------------------------------------------------------------------------------------------------------------------|
| (A)         | Democritus (460 BC) | A. Everything is made of atoms that cannot be created, divided, or destroyed.                                                      |
| (B)         | Bohr (1913)         | B. Most of an atom's mass is mostly concentrated in a small, dense nucleus, surrounded by a cloud of negatively charged electrons. |
| (C)         | Thomson (1897)      | C. "Solid Indivisible (continuous)"                                                                                                |
| (D)         | Rutherford (1911)   | D. Group Number (Valence Electrons)                                                                                                |
| (E)         | Moseley (1913)      | E. Atomic Mass                                                                                                                     |
| (F)         | Schrodinger (1926)  | F. Atoms mostly empty space, the negatively charged electrons move.                                                                |

**Directions:** Use the periodic table to answer the following questions:

- (100%)** (P.4.1.1) The number of protons equals the number of electrons.
- (100%)** (P.4.1.2) The atomic number will equal the **number of protons**.
- (75%)** (P.4.1.3) **(100%)** The number of protons in an atom is the same as the number of electrons in an atom.

# Electrons In Atoms Study Guide Key

**Daniela Niemeyer**



## **Electrons In Atoms Study Guide Key:**

**The Master Key System with Study Guide** Charles F. Haanel, 2022-06-28 This powerful beautifully designed book includes a 21st century study guide filled with practices and exercises that will flood your conscious and subconscious mind heart and soul with positive energy and life enhancing ideas The Master Key System one of the finest studies in personal power metaphysics and prosperity consciousness is possibly the greatest personal development book ever written Among the first books to use the phrase Law of Attraction which has become such a popular term today when referring to the most powerful force in the universe The Master Key System along with The Science of Getting Rich was the source of Rhonda Byrne s inspiration for the book and film The Secret Originally published in 1912 as a correspondence course it teaches that everything around us was first just a thought Using it as the sign of creative energy thought becomes a powerful force that can turn any idea into a reality In this deluxe edition and study guide you will learn How to use the power of thought to realize your dreams How to create abundance and wealth How to find happiness New Thought beliefs including The Law of Attraction Creative Visualization Ultimate principles causes effects and laws that underlie success The Importance of truth and harmony How to unlock your potential It is rumored that while attending Harvard University Bill Gates discovered The Master Key System It was this very book that inspired him to drop out and pursue his dream of a computer on every desktop

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atom the author explains molecules electronic configuration bonding hydrocarbons polar reaction mechanisms stereochemistry reaction varieties organic spectroscopy aromaticity and aromatic reactions biomolecules organic polymers and a synthetic approach to organic compounds The over one hundred diagrams and charts contained in this volume will help students visualize the structures and bonds as they read the text and make the logic of organic chemistry clear and easily understood Each chapter ends with a list of frequently asked questions and answers followed by additional practice problems Answers are included in the Appendix      *Organic Chemistry, 5e Student Study Guide and Solutions Manual* David R. Klein, Laurie S. Starkey, 2025-03-18 Success in organic chemistry requires mastery in two core aspects fundamental concepts and the skills needed to apply those concepts and solve problems With *Organic Chemistry Student Study Guide and Solutions Manual* 5th Edition students can learn to become proficient at approaching new situations methodically based on a repertoire of skills These skills are vital for successful problem solving in organic chemistry

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