

Atoms: Protons, Neutrons & Electrons Worksheets



Atoms: Protons, Neutrons, and Electrons

At the smallest scale of matter that still retains chemical properties, the nucleus contains positive protons and neutral neutrons. Surrounding the nucleus is a cloud of negative electrons that occupy specific energy levels. The number of protons in an atom determines the type of atom. The number of electrons explains the reactivity of an atom, influencing stability and charge. Understanding the structure of atoms is important in understanding matter and how it interacts.

Parts of an Atom

Fill in the Table According to each Particle

Particle	Name	Charge	Location

Unknown Atom

Fill in the Table According to Unknown Atom

Atom Name	
Atomic Number	
Proton Number	
Neutron Number	
Electron Number	

Atoms on the Periodic Table

Use the Periodic Table to Answer the Questions about Atoms

What is the unknown atom's group number?	Look at the periodic table.
What is the unknown atom's family number?	Look at the periodic table.

Elements on the Periodic Table

Fill in the Parts on the Elemental Square

Practice the Structure of the Elements on the Periodic Table

20
Ca
Calcium
40.078

- Number of Protons (and Electrons)
- Element Abbreviation
- Period Element Name
- Equal to Protons + Neutrons

How to Find Protons? → Look at the Atomic Number

How to Find Electrons? → Look at the Atomic Number

How to Find Neutrons? → Take Mass # - Atomic #

Mass # _____ Atomic # _____ Neutrons # _____

Elemental Square

Fill in the Elemental Square for Each Atom

12
Mg
Magnesium
24.305

Atom's Name
Atomic #
Proton #
Electron #
Neutron #

27
Co
Cobalt
58.933

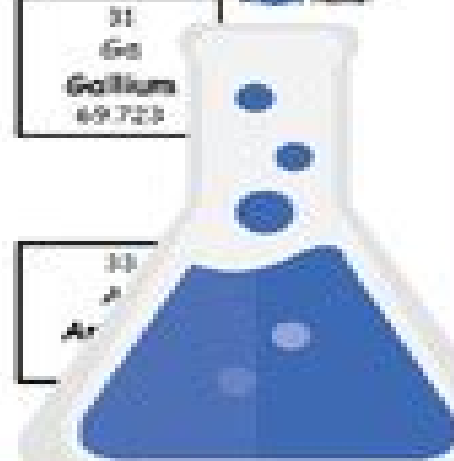
Atom's Name
Atomic #
Proton #
Electron #
Neutron #

31
Ga
Gallium
69.723

Atom's Name
Atomic #
Proton #
Electron #
Neutron #

33
As
Arsenic
74.922

Atom's Name
Atomic #
Proton #
Electron #
Neutron #



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M Walker



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