## 8 • Electron Configurations & Periodicity

#### WRITING ELECTRON CONFIGURATIONS

For each given element, fill in the orbital diagram and then write the electron configuration for the element.

1.	2.	3.	4.	5.	6.
	, <u>S</u>				S S S S S S S S S S S S S S S S S S S
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¹	¹⊠	180	180	'₩	'₩
Element: Ar # of e 's: 16	Element: Mg # of e-'s: 12	Element: N # of e-'s: 7	Element: Li # of e-'s: _3	Element: P # of e 's: 15	Element: Cl # of e <sup>-</sup> 's: 17

Write the electron configurations of each of these in long form and short form:

- Ar 152 252 206 352 306
  - AT TNET 352 3pt
- Mg 152 252 206 352
  - Mg [Ne] 352
- N 152 252 2p3 N [He] 252 2p3
- Li 152 25'
- P 152 252 Zpc 352 3p3
  - P [Ne] 35+ 3p3
- CI 152 252 2p6 352 3p5
- CI [Ne] 352 305



Fill in the orbital diagram for the element, Fe, and write the electron configuration of Fe in the long and short form.

Fe 1s2 252 2p 352 3p6 3d6 452

Fe [Ar] 3d 452



# **Electron Configuration And Periodicity Lab Key**

**Chao Zhang** 

#### **Electron Configuration And Periodicity Lab Key:**

The History and Use of Our Earth's Chemical Elements Robert E. Krebs, 2006-07-30 Understanding the Periodic Table of Chemical Elements is critical for success in the chemistry classroom and laboratory In today s classroom students not only need to understand the properties of the chemical elements but how these elements play such an integral role in industry the earth and the environment and in modern life No resource provides a better introduction than Robert Krebs s The History and Use of Our Earth's Chemical Elements In this thoroughly revised edition with extensive new examples on the importance of the chemical elements the elements are examined within their groups enabling students to make connections between elements of similar structure In addition the discovery and history of each element from those known from ancient times to those created in the modern laboratory is explained clearly and concisely Understanding the Periodic Table of Chemical Elements is critical for success in the chemistry classroom and laboratory In today's classroom students not only need to understand the properties of the chemical elements but how these elements play such an integral role in industry the earth and the environment and in modern life No resource provides a better introduction than Robert Krebs s The History and Use of Our Earth's Chemical Elements In this thoroughly revised edition with extensive new and updated examples on the use of the chemical elements the elements are examined within their groups enabling students to make connections between elements of similar structure In addition the discovery and history of each element from those known from ancient times to those created in the modern laboratory is explained clearly and concisely In addition to the handy Guide to the Chemical Elements that comprises the bulk of the work The History and Use of Our Earth's Chemical Elements includes other useful features Introductory material on the basics of chemistry and the Periodic Table Appendices on the discoverers of the chemical elements A glossary of words commonly used in chemistry and chemical engineering A complete bibliography of useful resources including websites All of this information makes The History and Use of Our Earth's Chemical Elements the ideal one volume resource for understanding the importance of the chemical elements Labster Virtual Lab Experiments: Basic Biochemistry Aaron Gardner, Wilko Duprez, Sarah Stauffer, Dewi Ayu Kencana Ungu, Frederik Clauson-Kaas, 2019-04-01 This textbook helps you to prepare for your next exams and practical courses by combining theory with virtual lab simulations The Labster Virtual Lab Experiments series gives you a unique opportunity to apply your newly acquired knowledge in a learning game that simulates exciting laboratory experiments Try out different techniques and work with machines that you otherwise wouldn t have access to In this book you ll learn the fundamental concepts of basic biochemistry focusing on Ionic and Covalent Bonds Introduction to Biological Macromolecules Carbohydrates Enzyme Kinetics In each chapter you ll be introduced to one virtual lab simulation and a true to life challenge Following a theory section you ll be able to play the relevant simulation that includes quiz questions to reinforce your understanding of the covered topics 3D animations will show you molecular processes not otherwise visible to the human eye If you have

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