# Potentiometry

# Choose the correct answer:

No	Sentence	answer
1	All of the following are related to reference electrode except:  a) of known and constant potential b) obeys Nernst equation. c) of high resistance to permit current of electric current. d) insensitive to activity of analyte solution.	с
2	Potential of standard hydrogen electrode is: a) Zero b) related to H* concentration c) 0.059 PH d) A and B e) All of the above	d
3	The increase in chloride ion concentration in Ag/AgCl electrode leads to: a) increase the potential b) decrease the potential c) no effect d) none of them	Ь
4	Potential of calomel electrode is proportional to chloride ion concentration.  a) directly b) inversely c) equal d) no relation	ь
5	In silver/silver chloride electrode, as KCl concentration increases:  a) silver ions increases, potential increases. b) silver ions decreases, potential increases. c) silver ions decreases, potential remains the same. d) silver ions decreases, potential decreases.	d
6	In calomel electrode, as KCI concentration increases:  a) mercury(I) ions increases, potential increases. b) mercury(I) ions decreases, potential increases. c) mercury(I) ions decreases, potential remains the same. d) mercury(I) ions decreases, potential decreases.	d
7	What is the electrode potential for a half-cell consisting of a cadmium electrode immersed in a solution that is 0.0150 M in Cd2+ (E° = -0.403 V) a) 0.457 b) 0.403 C) -0.457 d) -0.201	c

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**Chandresh Agrawal** 

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web sahih muslim 804 a abu umama said he heard allah s messenger [] say recite the qur an for on the day of resurrection it will come as an intercessor for those who recite it recite the two bright ones al baqara and surah al imran for on the day of resurrection they will come as two clouds or two shades or two flocks of birds in ranks

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