

(a)

Reactants	Products
H, Cl and HCl	2H, O ₂ and H ₂ O

(b)

Reactants	Products
HCl and 2H ₂ O	H, Cl, 2H and O ₂

(c)

Reactants	Products
H, Cl, 2H and O ₂	HCl and 2H ₂ O

(d)

Reactants	Products
2H, O ₂ and H ₂ O	H, Cl and HCl

Correct Answer: Option C

Q4) A student performs an experiment to form aluminium chloride from aluminium and chlorine. Which options give the chemical equation of the reaction?

- (a) $\text{Al} + \text{Cl}_2 \rightarrow \text{AlCl}_2$
 (b) $2\text{Al} + \text{Cl}_2 \rightarrow 2\text{AlCl}$
 (c) $2\text{Al} + 3\text{Cl}_2 \rightarrow 2\text{AlCl}_3$
 (d) $3\text{Al} + 3\text{Cl}_2 \rightarrow 3\text{AlCl}_3$

Correct Answer: Option C

Q5. Give the ratio in which hydrogen and oxygen are present in water by volume.

- (a) 1:2
 (b) 1:1
 (c) 2:1

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