

# INTRODUCTION TO MANUFACTURING PROCESSES

SECOND EDITION

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Professor of Mechanical Engineering  
University of California, Berkeley

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# Introduction To Manufacturing Processes Schey Solution Manual

**Julia Schneider**



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 $\text{N}_2 + 3\text{H}_2 \rightarrow 2\text{NH}_3$   
 $\text{Mg}(\text{OH})_2 + 2\text{HNO}_3 \rightarrow \text{Mg}(\text{NO}_3)_2 + 2\text{H}_2\text{O}$   
 $2\text{NaClO}_3 \rightarrow 2\text{NaCl} + 3\text{O}_2$   
 $4\text{Al} + 3\text{O}_2 \rightarrow 2\text{Al}_2\text{O}_3$   
 $\text{N}_2 + \text{G} \rightarrow 2\text{NH}_3$

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 $\text{Mg}(\text{OH})_2 + \text{HCl} \rightarrow \text{MgCl}_2 + \text{H}_2\text{O}$   
note all reactants and products require a coefficient of at least one  
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 $2\text{C}_2\text{H}_2 + 5\text{O}_2 \rightarrow 4\text{CO}_2 + 2\text{H}_2\text{O}$   
if 1.0 mol of  $\text{C}_2\text{H}_2$  is burned how many moles of  $\text{CO}_2$  are formed  
if 1.0 mol of  $\text{C}_2\text{H}_2$  is burned how many moles of oxygen gas are consumed

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water  
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zinc nitrate  
copper

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