

B Chemistry – 16

Topic 3 Questions

1. Which of the following salts forms coloured solutions when dissolved in water?

- i. NaCl_2
 - ii. FeCl_3
 - iii. NaCl_2
 - iv. ZnCl_2
- A. i and ii only
B. ii and iii only
C. iii and iv only
D. i, ii, iii and iv

[Total: 1 mark]

2. Which is an essential feature of a ligand?

- A. a negative charge
B. an odd number of electrons
C. the presence of two or more atoms
D. the presence of a non-bonding pair of electrons

[Total: 1 mark]

3. Which equation represents the third ionisation energy of an element M?

- A. $\text{M}^+(g) \rightarrow \text{M}^{2+}(g) + 3e^-$
B. $\text{M}^{2+}(g) \rightarrow \text{M}^{3+}(g) + e^-$
C. $\text{M}(g) \rightarrow \text{M}^{2+}(g) + 3e^-$
D. $\text{M}^{2+}(g) \rightarrow \text{M}^{3+}(g) + e^-$

[Total: 1 mark]

4. Which electrons are lost by an atom of iron when it forms the Fe^{2+} ion?

- A. One s orbital electron and two d orbital electrons
B. Two s orbital electrons and one d orbital electron
C. Three s orbital electrons
D. Three d orbital electrons

[Total: 1 mark]

5. Which properties are typical of d-block elements?

- i. complex ion formation

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