

Concept Review Chapter 5: Section 1

Section: Simple Ions

Complete each statement below by choosing a term from the following list. Terms may be used more than once.

10

11

ion

octet

12

13

anion

cation

1. An atom or group of atoms that has a positive or negative electric charge because it has lost or gained electrons is a(n) ion.
2. The sodium ion has 11 protons, 12 neutrons, and 10 electrons.
3. In most chemical reactions, atoms tend to match the outer electron configuration of the noble gases. This is called the octet rule.

Complete each statement below by writing the correct term or phrase.

4. An anion is an ion with a negative charge.
5. Valence electrons occupy the outermost energy level of an atom.
6. The chloride ion and the chlorine atom have different electron configurations.
7. A cation is an ion with a positive charge.
8. Many stable ions have an electron configuration of a noble gas.

Answer the following questions in the space provided.

9. How do the outer-shell electron configurations for ions of Group 1, Group 2, and Group 15, Group 16, and Group 17 elements compare with those of the noble gases?

They are the same. Group 1 + 2 ions have noble gas configuration of previous row. Group 15, 16 + 17 ions have noble gas configuration in their row.

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Table of Contents Holt Science Spectrum Matter Answer

1. Understanding the eBook Holt Science Spectrum Matter Answer
 - The Rise of Digital Reading Holt Science Spectrum Matter Answer
 - Advantages of eBooks Over Traditional Books
2. Identifying Holt Science Spectrum Matter Answer
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Holt Science Spectrum Matter Answer
 - User-Friendly Interface
4. Exploring eBook Recommendations from Holt Science Spectrum Matter Answer
 - Personalized Recommendations
 - Holt Science Spectrum Matter Answer User Reviews and Ratings
 - Holt Science Spectrum Matter Answer and Bestseller Lists
5. Accessing Holt Science Spectrum Matter Answer Free and Paid eBooks
 - Holt Science Spectrum Matter Answer Public Domain eBooks
 - Holt Science Spectrum Matter Answer eBook Subscription Services
 - Holt Science Spectrum Matter Answer Budget-Friendly Options
6. Navigating Holt Science Spectrum Matter Answer eBook Formats
 - ePub, PDF, MOBI, and More
 - Holt Science Spectrum Matter Answer Compatibility with Devices
 - Holt Science Spectrum Matter Answer Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Holt Science Spectrum Matter Answer
 - Highlighting and Note-Taking Holt Science Spectrum Matter Answer
 - Interactive Elements Holt Science Spectrum Matter Answer
8. Staying Engaged with Holt Science Spectrum Matter Answer

- Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Holt Science Spectrum Matter Answer
9. Balancing eBooks and Physical Books Holt Science Spectrum Matter Answer
- Benefits of a Digital Library
 - Creating a Diverse Reading Collection Holt Science Spectrum Matter Answer
10. Overcoming Reading Challenges
- Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Holt Science Spectrum Matter Answer
- Setting Reading Goals Holt Science Spectrum Matter Answer
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Holt Science Spectrum Matter Answer
- Fact-Checking eBook Content of Holt Science Spectrum Matter Answer
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
- Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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