

Name _____ date _____

Ideal Gas Law Worksheet Review

$$PV=nRT$$

mass \longleftrightarrow mole

$$R = \begin{array}{l} 0.0821 \text{ L}\cdot\text{atm/mol}\cdot\text{K} \\ 62.4 \text{ L}\cdot\text{mm Hg/mol}\cdot\text{K} \\ 8.314 \text{ L}\cdot\text{kPa/mol}\cdot\text{K} \end{array}$$

Standard pressure and temperature (STP): 1 atm and 273 K

1. How many moles of a gas at 100°C does it take to fill a 1.00 L flask to a pressure of 1.50 atm?
2. A camping stove holds 3000 g of liquid C_3H_8 . How large of a container would be needed to hold the gas at a temperature of at 25°C and a pressure of 3.0 atm? (hint: need to convert to moles first)
3. What volume would be occupied by 3.5 mol of oxygen gas at a pressure of 768 mm Hg and a temperature of 25°C?
4. A ten-liter gas cylinder contains 55 moles nitrogen. What pressure, in kPa, is exerted by the nitrogen at 25°C?
5. A drum used to transport crude oil has a volume of 162 L. How many grams of water, as steam, are required to fill the drum at 1.20 atm and 100°C?
6. How many moles of air are there in a 125 mL flask if the pressure is 755 mm Hg and the temperature is 20°C?

Ideal Gas Law Worksheet With Answers

Guillaume Favre

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

Ideal Gas Law Worksheet With Answers:

Decoding **Ideal Gas Law Worksheet With Answers**: Revealing the Captivating Potential of Verbal Expression

In a period characterized by interconnectedness and an insatiable thirst for knowledge, the captivating potential of verbal expression has emerged as a formidable force. Its ability to evoke sentiments, stimulate introspection, and incite profound transformations is genuinely awe-inspiring. Within the pages of "**Ideal Gas Law Worksheet With Answers**," a mesmerizing literary creation penned with a celebrated wordsmith, readers set about an enlightening odyssey, unraveling the intricate significance of language and its enduring impact on our lives. In this appraisal, we shall explore the book's central themes, evaluate its distinctive writing style, and gauge its pervasive influence on the hearts and minds of its readership.

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