#### 18. Lattice enthalpies at 298 K (experimental values)

The lattice enthalpy values  $(\Delta H^a_{\rm lettice})$  given relate to the endothermic process  $M_a X_b(s) \to a M^{b+}(g) + b X^{a-}(g)$  in which the gaseous ions of a crystal are separated to an infinite distance from each other.

#### Experimental values

The data in these tables are experimental values obtained by means of a suitable Born-Haber cycle.

Alkali metal halides	$\Delta H^{\nu}_{\text{lattice}}$ (kj med <sup>-1</sup> )			
	P	а	Be	1
Lii	1049	864	820	764
Na	930	790	754	705
K	829	720	691	650
Rb	795	695	668	632
Cs	759	670	647	613

Other substances	ΔH <sup>o</sup> <sub>lattion</sub> (kl mol <sup>-1</sup> )		
CaF <sub>2</sub>	2651		
BeCl <sub>2</sub>	3033		
MgCl <sub>3</sub>	2540		
CaClg	2271		
SrCl <sub>2</sub>	2170		
BaClg	2069		
MgO	3791		
CaO	3401		

Other substances	ΔN <sub>Lettice</sub> (k) mol <sup>-1</sup> )
SrO	3223
BaO	3054
OrCl <sub>2</sub>	2824
AgF	974
AgCI	918
AgBr	905
Agt	892

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**Kathleen Armour** 

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