Periodic Table Named-Groups Worksheet – 2

DIRECTIONS: Use the Periodic Table shown below to answer each question, <u>using the letters in</u> each "box" to identify the required element(s).

A		-										- 1
									D		E	
C												
	E				G			н				- K
	L	П								I		Г
J		Z		M			N				0	F
o		\R\		s	т	0						
		/	\					<u> </u>				2
		1	v*I				w					
			×				Y					

List elements in alphabetical order when more than 1 element answers a question.

							C) it is NOT a
gas I	D) it does NOT	share the pr	operties of the	e other elemen	nts in its group	E) it is a liqu	bis

Which of these	elements is/are	highly reactive?	A)	B)	C)	D) A, B, and C	E) A and B only

3) What is the letter of the element whose atomic number is closest to the element labeled "x"?

4) Which letter labels the halogen element with the lowest atomic number?

5) Which of these elements is the least reactive? A) "o" B) "Y" C) "A" D) "o" E) "B"

6) Which of these elements is/are NOT transition elements/metals (list ALL that are not transition elements)? A) "c" B) "r" C) "r" D) "r" E) "r"

continued on other side

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M Woodhall

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