SNAP

: Balance the following chemical equations

20.
$$AI + CI_2 \rightarrow AICI_3$$

21. $AI + FeO \rightarrow AI_2O_3 + Fe$
22. $Na_2CO_3 + HCI \rightarrow H_2CO_3 + NaCI$
23. $P_4O_{10} + P_4O \rightarrow H_3PO_4$
24. $AI_2(SO_4)_3 + Ca(OH)_2 \rightarrow AI(OH)_3 + Ca$
25. $C_3H_8 + O_2 \rightarrow CO_2 + H_2O$
26. $Fe_2O_3 + H_2 \rightarrow Fe + H_2O$
27. $N_2O_5 + H_2O \rightarrow HNO_3$
28. $H_3AsO_4 \rightarrow As_2O_5 + H_2O$
29. $Pb + Na + C_3H_3CI \rightarrow Pb(C_3H_5)_4 + Ca$

Part IV: In each of the following problems you are given some potent reaction. Determine what type of reaction would occur, predict the p the complete chemical equation and balance it!

- 30. Aluminum powder burns in the presence of oxygen to form ...
- 31. Magnesium metal reacts with nitric acid to form ...
- 32. Aqueous barium chloride reacts with sodium hydroxide to form ...
- 33. Butane burns in the presence of oxygen to form ...
- 34. Hydrogen gas reacts with chlorine gas to form ...
- Aluminum powder reacts with iron(III) oxide to form...

Formulas And Chemical Reactions Study Guide

Zhe Xu

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