

Forensics Rubric and Answer Key

(Total: 278 points, TIEBREAKER: 6pts)

Princeton Science Olympiad 2017

I. Qualitative Analysis (42 pts)

You are given 12 samples of white powders. Use any method available to you to determine their identities, and then complete the following table, indicating each unknown's name and chemical formula.

No double jeopardy here, if name is wrong but formula matches, +1

| Powder | Name of Powder (1pts each) | Chemical Formula (1pts each) |
|--------|---|---|
| A | Sodium hydrogen carbonate/ bicarbonate | NaHCO_3 |
| B | Lithium chloride | LiCl |
| C | glucose | $\text{C}_6\text{H}_{12}\text{O}_6$ |
| D | Sodium chloride | NaCl |
| E | sucrose | $\text{C}_{12}\text{H}_{22}\text{O}_{11}$ |
| F | Sodium acetate | $\text{NaC}_2\text{H}_3\text{O}_2$ |
| G | cornstarch | $\text{C}_{27}\text{H}_{48}\text{O}_{20}$ |
| H | Magnesium sulfate | MgSO_4 |
| I | Calcium nitrate | $\text{Ca}(\text{NO}_3)_2$ |
| J | Potassium chloride | KCl |
| K | Ammonium chloride | NH_4Cl |
| L | Calcium sulfate | CaSO_4 |

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